

Alkyne Hydroamination and Trimerization with Titanium *Bis*(phenolate)pyridine Complexes: Evidence for Low-Valent Titanium Intermediates and Synthesis of an Ethylene Adduct of Ti(II)

Ian A. Tonks, Josef C. Meier and John E. Bercaw*

Contribution from the Arnold and Mabel Beckman Laboratories of Chemical Synthesis, California Institute of Technology, Pasadena, California 91125. Received xxxxxxxx xx, 2013.
E-mail: bercaw@caltech.edu

Supporting Information

Contents

Figure S1. ^1H NMR spectrum of 2 (C_6D_6).....	2
Figure S2. ^1H NMR spectrum of 3 (C_6D_6).....	3
Figure S3. ^1H NMR spectrum of 4 (C_6D_6).....	4
Figure S4. ^1H NMR spectrum of 6 (C_6D_6).....	5
Table S1. Crystal and refinement data for complexes 2 , 4 , 6 and 7	6
Tables of bond lengths, angles, and anisotropic displacement parameters	
Complex 2.....	7
Complex 4.....	20
Complex 6.....	32
Complex 7.....	48

Figure S1. ^1H NMR spectrum of **2** (C_6D_6).

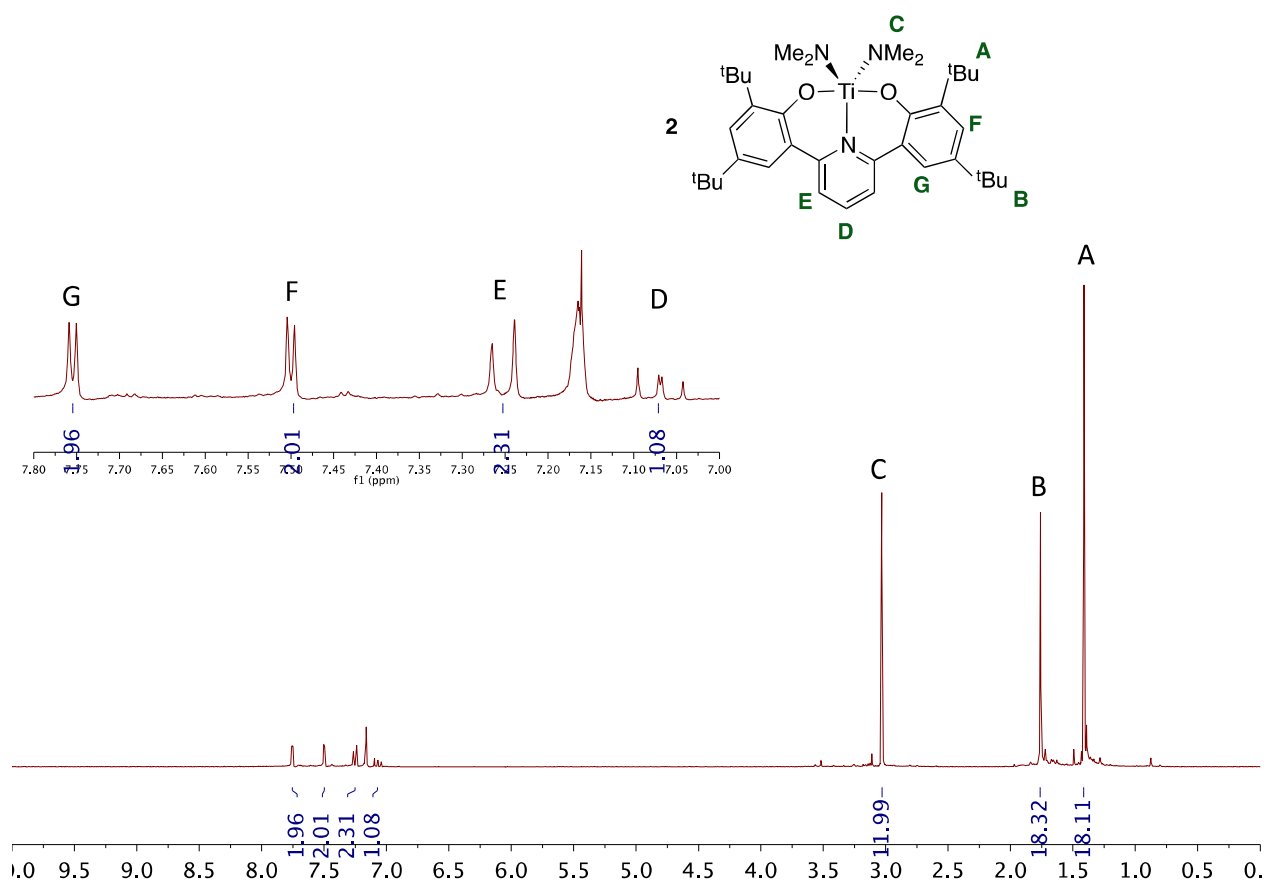


Figure S2. ^1H NMR spectrum of **3** (C_6D_6).

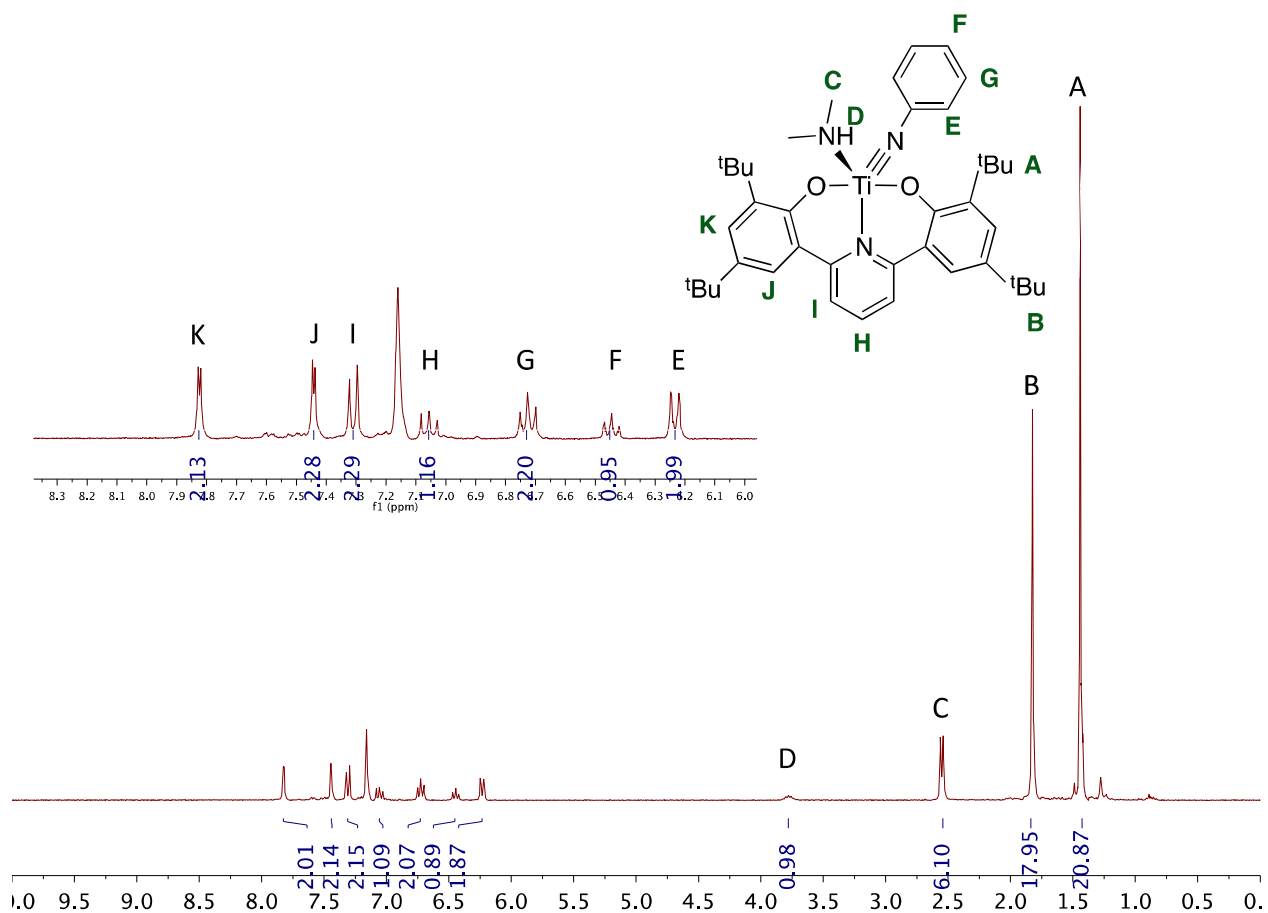


Figure S3. ^1H NMR spectrum of **4** (C_6D_6).

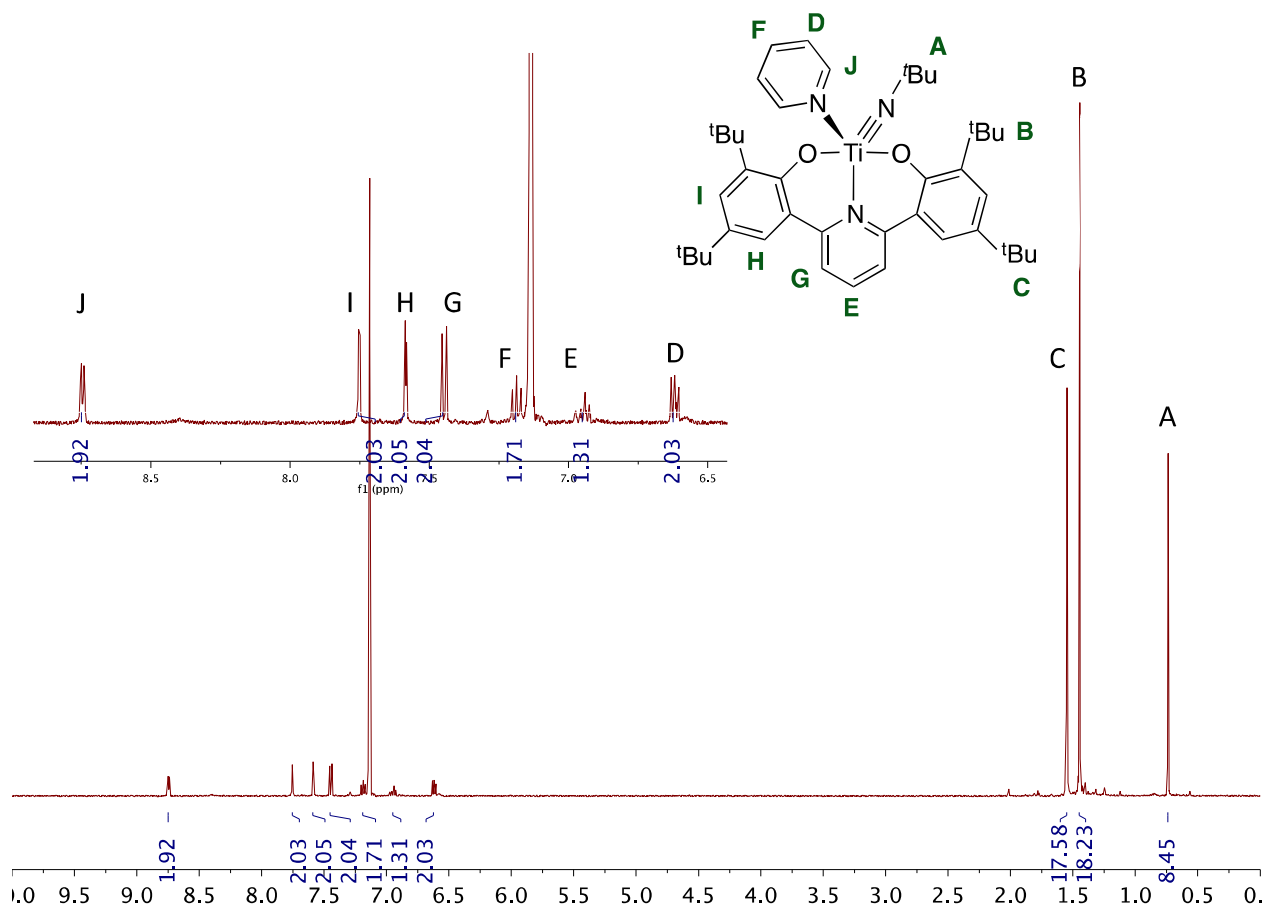


Figure S4. ^1H NMR spectrum of **6** (C_6D_6).

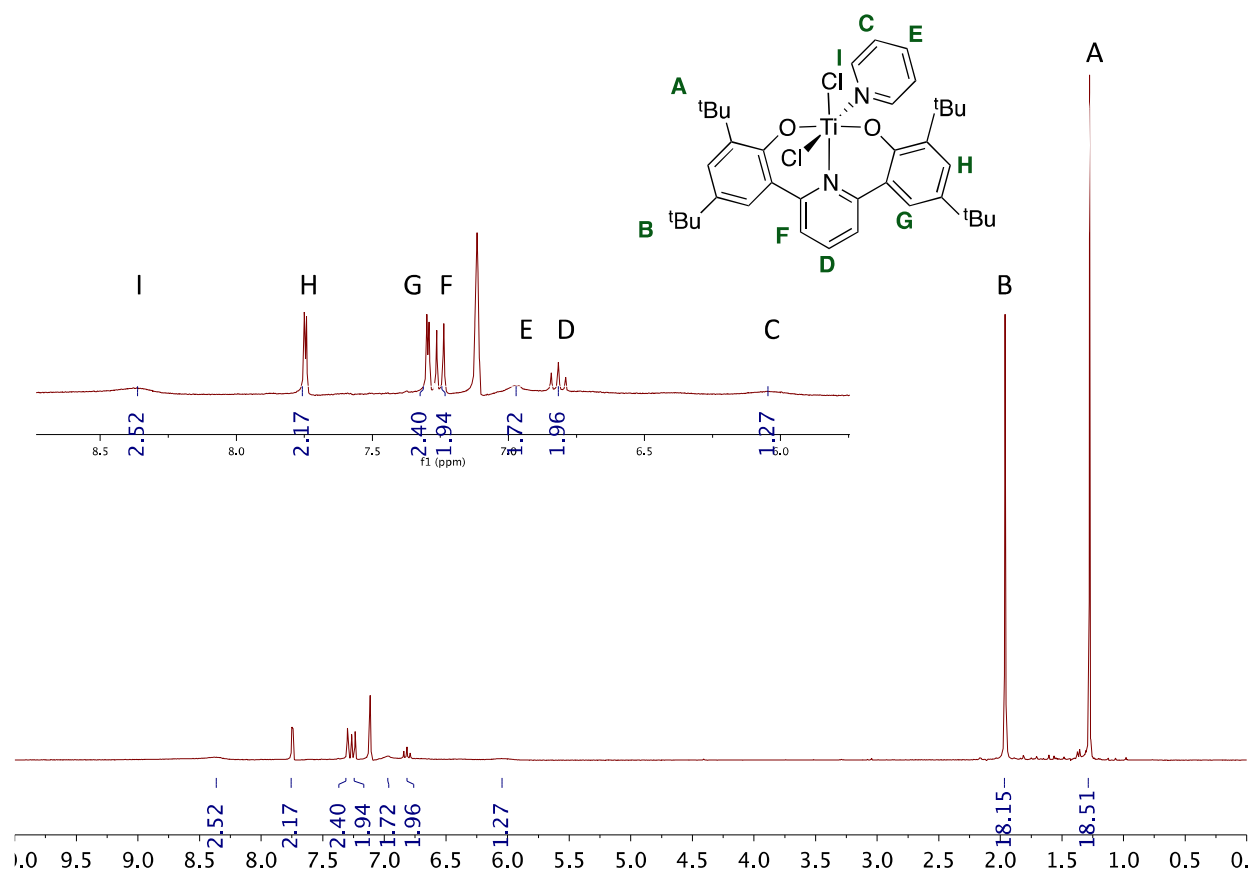


Table S1. Crystal and refinement data for complexes **2**, **4**, **6**, and **7**.

	2	4	6	7
CCDC Number	754732	847140	856446	822994
Empirical formula	C ₃₇ H ₅₅ N ₃ O ₂ Ti • C ₅ H ₁₂	C ₄₂ H ₅₇ N ₃ O ₂ Ti • 0.49(C ₅ H ₁₂) • 0.51(C ₇ H ₈)	2(C ₃₈ H ₄₈ N ₂ O ₂ Cl ₂ Ti) • 5(C ₆ H ₆)	C ₃₇ H ₅₃ N ₂ O ₂ Ti • 0.5(C ₅ H ₁₂)
Formula weight	693.89	765.95	878.85	641.79
T (K)	100(2)	100(2)	100(2)	100(2)
<i>a</i> , Å	13.9103(6)	14.6930(8)	11.8721(6)	17.7536(5)
<i>b</i> , Å	10.2228(4)	14.9636(8)	28.5845(14)	10.5239(3)
<i>c</i> , Å	28.5988(12)	20.1744(12)	29.6698(15)	20.7880(6)
α , deg				
β , deg	90.995(3)	92.699(3)	100.018(3)	104.343(2)
γ , deg				
Volume, Å ³	4066.2(3)	4430.6(4)	9915.2(9)	3762.91(19)
<i>Z</i>	4	4	8	4
Crystal system	Monoclinic	Monoclinic	Monoclinic	Monoclinic
Space group	P 2 ₁ / <i>n</i>	P 2 ₁ / <i>c</i>	P 2 ₁ / <i>c</i>	P 2 ₁ / <i>n</i>
<i>d</i> _{calc} , g/cm ³	1.133	1.148	1.177	1.133
θ range, deg	1.62 to 29.10	1.70 to 30.51	1.59 to 35.13	2.02 to 29.99
μ , mm ⁻¹	0.247	0.233	0.320	0.261
Abs. Correction	None	None	None	None
GOF	1.764	1.945	1.877	2.095
<i>R</i> ₁ , ^a	R1 = 0.0650,	R1 = 0.0584,	R1 = 0.0415,	R1 = 0.0612,
<i>wR</i> ₂ , ^b [<i>I</i> > 2 σ (<i>I</i>)]	<i>wR</i> 2 = 0.0942	<i>wR</i> 2 = 0.0723	<i>wR</i> 2 = 0.0625	<i>wR</i> 2 = 0.0614

$$^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|. \quad ^b wR_2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]]^{1/2}.$$

CALIFORNIA INSTITUTE OF TECHNOLOGY
BECKMAN INSTITUTE
X-RAY CRYSTALLOGRAPHY LABORATORY



Date 13 November 2009

Crystal Structure Analysis of:
Complex 2 (ONO)Ti(NMe₂)₂ (IAT26)
(shown below)

By Michael W. Day 116 Beckman ext. 2734
e-mail: mikeday@caltech.edu

Contents

Table 1. Crystal data

Figures Minimum overlap, unit cell contents

Table 2. Atomic Coordinates

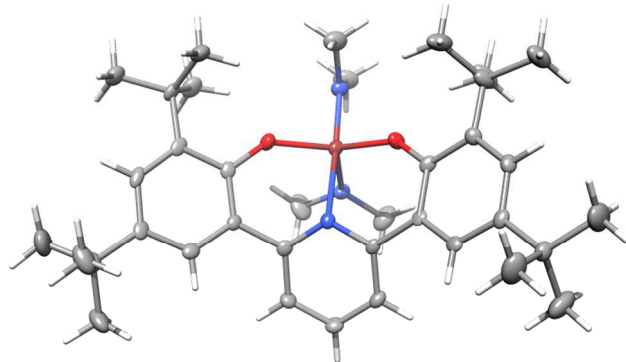
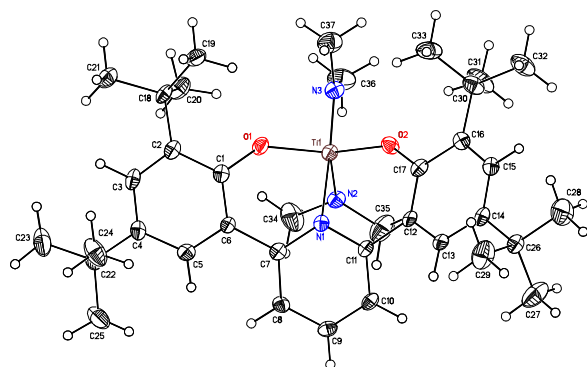
Table 3. Selected bond distances and angles

Table 4. Full bond distances and angles

Table 5. Anisotropic displacement parameters

Table 6. Hydrogen atomic coordinates

Table 7. Observed and calculated structure factors (available upon request)



IAT26

Note: The crystallographic data have been deposited in the Cambridge Database (CCDC). The deposition number is 754732.

Table 1. Crystal data and structure refinement for IAT26 (CCDC 754732).

Empirical formula	$C_{37}H_{55}N_3O_2Ti \cdot C_5H_{12}$
Formula weight	693.89
Crystallization Solvent	Pentane
Crystal Habit	Column
Crystal size	0.39 x 0.14 x 0.09 mm ³
Crystal color	Yellow



Data Collection

Type of diffractometer	Bruker KAPPA APEX II	
Wavelength	0.71073 Å MoK α	
Data Collection Temperature	100(2) K	
θ range for 9954 reflections used in lattice determination	2.45 to 24.33°	
Unit cell dimensions	a = 13.9103(6) Å b = 10.2228(4) Å c = 28.5988(12) Å	$\alpha = 90^\circ$ $\beta = 90.995(3)^\circ$ $\gamma = 90^\circ$
Volume	4066.2(3) Å ³	
Z	4	
Crystal system	Monoclinic	
Space group	P 2 ₁ /n	
Density (calculated)	1.133 Mg/m ³	
F(000)	1512	
Data collection program	Bruker APEX2 v2009.7-0	
θ range for data collection	1.62 to 29.10°	
Completeness to $\theta = 29.10^\circ$	90.2 %	
Index ranges	-17 \leq h \leq 18, -13 \leq k \leq 13, -38 \leq l \leq 36	
Data collection scan type	ω scans; 8 settings	
Data reduction program	Bruker SAINT-Plus v7.66A	
Reflections collected	51468	
Independent reflections	9823 [$R_{int} = 0.0899$]	
Absorption coefficient	0.247 mm ⁻¹	
Absorption correction	None	
Max. and min. transmission	0.9781 and 0.9099	

Table 1 (cont.)**Structure solution and Refinement**

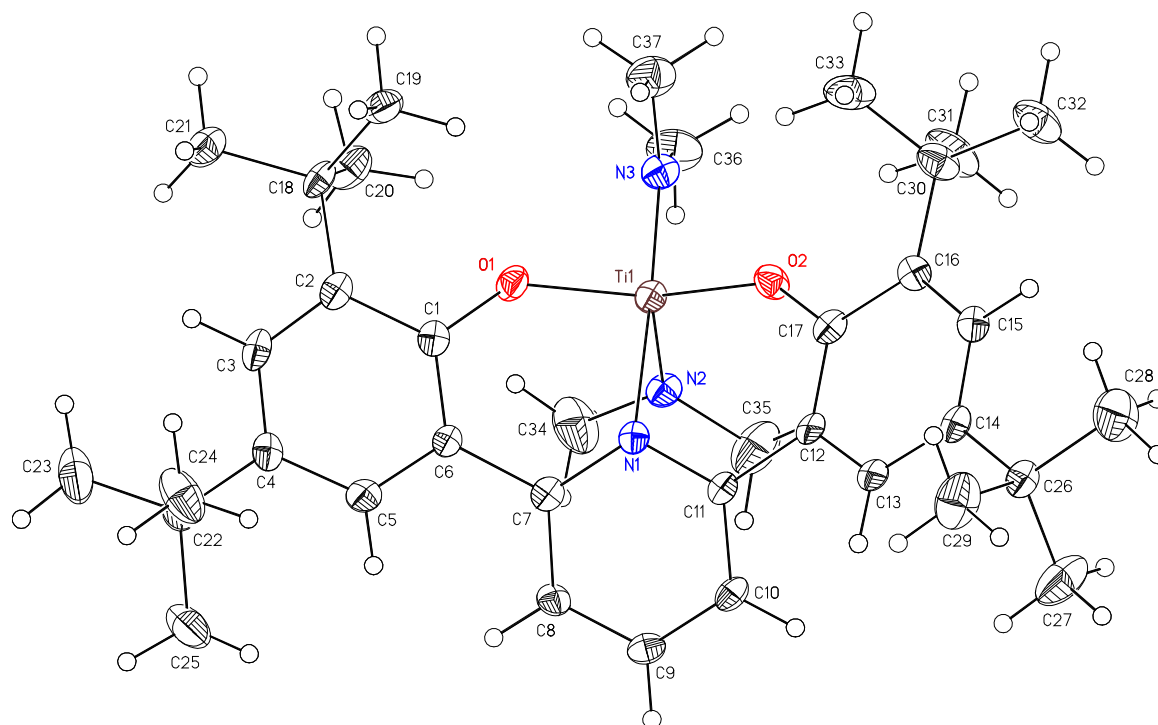
Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on F^2
Data / restraints / parameters	9823 / 0 / 451
Treatment of hydrogen atoms	Riding
Goodness-of-fit on F^2	1.764
Final R indices [$I > 2\sigma(I)$, 6111 reflections]	$R1 = 0.0650$, $wR2 = 0.0942$
R indices (all data)	$R1 = 0.1168$, $wR2 = 0.0977$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.001
Average shift/error	0.000
Largest diff. peak and hole	0.895 and -0.736 e.Å ⁻³

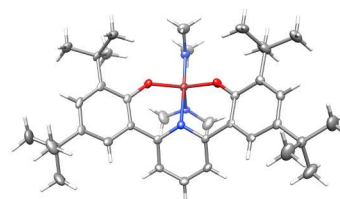
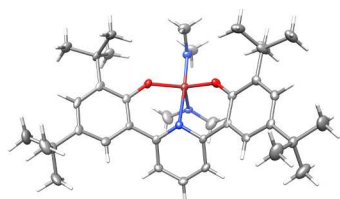
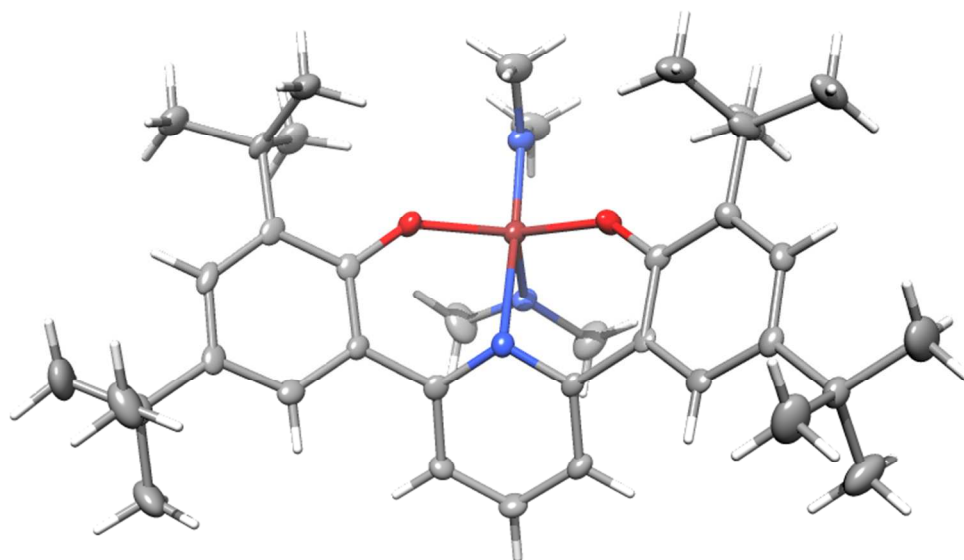
Special Refinement Details

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

Refinement of F^2 against ALL reflections. The weighted R-factor (wR) and goodness of fit (S) are based on F^2 , conventional R-factors (R) are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.





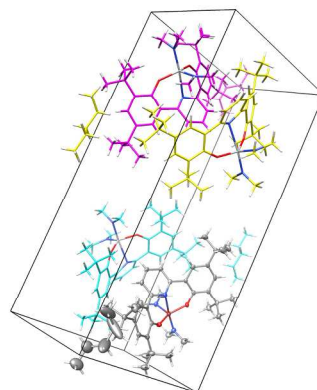
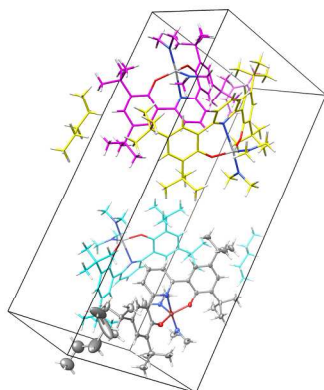
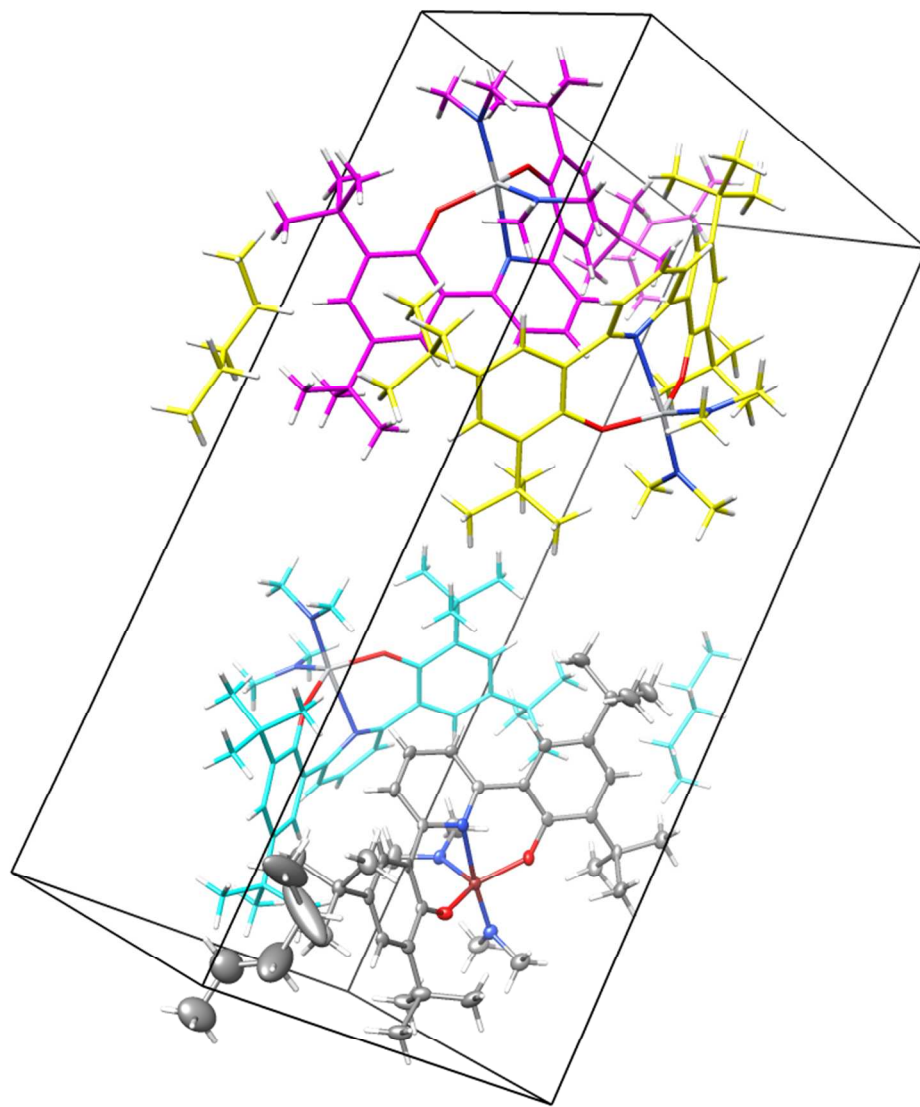


Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for IAT26 (CCDC 754732). $U(\text{eq})$ is defined as the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U_{eq}
Ti(1)	9254(1)	4380(1)	8263(1)	17(1)
O(1)	9890(1)	5170(1)	7765(1)	22(1)
O(2)	8338(1)	5010(1)	8677(1)	22(1)
N(1)	7928(1)	4731(2)	7734(1)	16(1)
N(2)	9142(1)	2584(2)	8154(1)	19(1)
N(3)	10320(1)	4386(2)	8687(1)	22(1)
C(1)	9782(2)	5404(2)	7303(1)	16(1)
C(2)	10533(2)	6045(2)	7065(1)	18(1)
C(3)	10394(2)	6281(2)	6589(1)	20(1)
C(4)	9577(2)	5907(2)	6335(1)	19(1)
C(5)	8873(2)	5250(2)	6581(1)	19(1)
C(6)	8959(2)	4986(2)	7060(1)	15(1)
C(7)	8137(2)	4356(2)	7290(1)	16(1)
C(8)	7549(2)	3461(2)	7048(1)	19(1)
C(9)	6724(2)	3020(2)	7250(1)	21(1)
C(10)	6451(2)	3521(2)	7677(1)	19(1)
C(11)	7064(2)	4398(2)	7908(1)	14(1)
C(12)	6754(2)	5039(2)	8349(1)	15(1)
C(13)	5790(2)	5375(2)	8392(1)	16(1)
C(14)	5445(2)	5991(2)	8789(1)	17(1)
C(15)	6112(2)	6271(2)	9141(1)	18(1)
C(16)	7081(2)	5971(2)	9121(1)	18(1)
C(17)	7408(2)	5352(2)	8710(1)	16(1)
C(18)	11466(2)	6444(2)	7326(1)	20(1)
C(19)	11243(2)	7497(2)	7692(1)	28(1)
C(20)	11927(2)	5242(2)	7561(1)	32(1)
C(21)	12213(2)	7018(2)	6992(1)	29(1)
C(22)	9441(2)	6244(2)	5816(1)	22(1)
C(23)	10396(2)	6296(3)	5569(1)	46(1)
C(24)	8957(2)	7588(2)	5771(1)	40(1)
C(25)	8806(2)	5239(2)	5567(1)	40(1)
C(26)	4373(2)	6340(2)	8820(1)	21(1)
C(27)	3761(2)	5127(2)	8731(1)	45(1)
C(28)	4112(2)	6883(3)	9300(1)	45(1)
C(29)	4118(2)	7370(2)	8448(1)	42(1)
C(30)	7780(2)	6311(2)	9526(1)	26(1)
C(31)	8329(2)	5085(2)	9697(1)	40(1)
C(32)	7261(2)	6878(2)	9949(1)	35(1)
C(33)	8490(2)	7355(2)	9357(1)	37(1)
C(34)	9678(2)	1938(2)	7789(1)	44(1)
C(35)	8424(2)	1745(2)	8354(1)	42(1)
C(36)	10909(2)	3323(2)	8885(1)	38(1)
C(37)	10666(2)	5638(2)	8854(1)	36(1)
C(41)	1042(3)	8522(3)	10329(1)	88(1)
C(42)	1231(3)	8503(3)	9827(1)	87(1)
C(43)	1618(4)	9583(5)	9598(2)	145(2)
C(44)	1816(3)	9515(6)	9079(2)	190(4)
C(45)	981(4)	9457(3)	8790(1)	135(2)

Table 3. Selected bond lengths [Å] and angles [°] for IAT26 (CCDC 754732).

Ti(1)-N(2)	1.8683(17)	N(2)-Ti(1)-O(2)	112.81(8)
Ti(1)-O(2)	1.8688(15)	N(2)-Ti(1)-O(1)	109.65(8)
Ti(1)-O(1)	1.8727(15)	O(2)-Ti(1)-O(1)	132.04(7)
Ti(1)-N(3)	1.8986(19)	N(2)-Ti(1)-N(3)	99.88(8)
Ti(1)-N(1)	2.3917(18)	O(2)-Ti(1)-N(3)	97.38(7)
		O(1)-Ti(1)-N(3)	96.33(8)
		N(2)-Ti(1)-N(1)	88.90(7)
		O(2)-Ti(1)-N(1)	79.71(7)
		O(1)-Ti(1)-N(1)	79.85(7)
		N(3)-Ti(1)-N(1)	171.19(7)

Table 4. Bond lengths [Å] and angles [°] for IAT26 (CCDC 754732).

Ti(1)-N(2)	1.8683(17)		
Ti(1)-O(2)	1.8688(15)	N(2)-Ti(1)-O(2)	112.81(8)
Ti(1)-O(1)	1.8727(15)	N(2)-Ti(1)-O(1)	109.65(8)
Ti(1)-N(3)	1.8986(19)	O(2)-Ti(1)-O(1)	132.04(7)
Ti(1)-N(1)	2.3917(18)	N(2)-Ti(1)-N(3)	99.88(8)
O(1)-C(1)	1.350(3)	O(2)-Ti(1)-N(3)	97.38(7)
O(2)-C(17)	1.345(2)	O(1)-Ti(1)-N(3)	96.33(8)
N(1)-C(11)	1.353(3)	N(2)-Ti(1)-N(1)	88.90(7)
N(1)-C(7)	1.364(3)	O(2)-Ti(1)-N(1)	79.71(7)
N(2)-C(35)	1.442(3)	O(1)-Ti(1)-N(1)	79.85(7)
N(2)-C(34)	1.454(3)	N(3)-Ti(1)-N(1)	171.19(7)
N(3)-C(37)	1.446(3)	C(1)-O(1)-Ti(1)	140.83(15)
N(3)-C(36)	1.469(3)	C(17)-O(2)-Ti(1)	143.37(15)
C(1)-C(6)	1.396(3)	C(11)-N(1)-C(7)	118.43(19)
C(1)-C(2)	1.417(3)	C(11)-N(1)-Ti(1)	114.26(14)
C(2)-C(3)	1.394(3)	C(7)-N(1)-Ti(1)	111.91(14)
C(2)-C(18)	1.540(3)	C(35)-N(2)-C(34)	112.50(19)
C(3)-C(4)	1.391(3)	C(35)-N(2)-Ti(1)	125.09(16)
C(4)-C(5)	1.388(3)	C(34)-N(2)-Ti(1)	121.59(16)
C(4)-C(22)	1.533(3)	C(37)-N(3)-C(36)	110.33(19)
C(5)-C(6)	1.399(3)	C(37)-N(3)-Ti(1)	117.76(15)
C(6)-C(7)	1.478(3)	C(36)-N(3)-Ti(1)	131.90(15)
C(7)-C(8)	1.401(3)	O(1)-C(1)-C(6)	120.7(2)
C(8)-C(9)	1.369(3)	O(1)-C(1)-C(2)	118.7(2)
C(9)-C(10)	1.383(3)	C(6)-C(1)-C(2)	120.5(2)
C(10)-C(11)	1.396(3)	C(3)-C(2)-C(1)	117.2(2)
C(11)-C(12)	1.491(3)	C(3)-C(2)-C(18)	122.0(2)
C(12)-C(13)	1.391(3)	C(1)-C(2)-C(18)	120.8(2)
C(12)-C(17)	1.403(3)	C(4)-C(3)-C(2)	124.1(2)
C(13)-C(14)	1.391(3)	C(5)-C(4)-C(3)	116.5(2)
C(14)-C(15)	1.386(3)	C(5)-C(4)-C(22)	121.5(2)
C(14)-C(26)	1.537(3)	C(3)-C(4)-C(22)	121.9(2)
C(15)-C(16)	1.385(3)	C(4)-C(5)-C(6)	122.7(2)
C(16)-C(17)	1.416(3)	C(1)-C(6)-C(5)	119.0(2)
C(16)-C(30)	1.539(3)	C(1)-C(6)-C(7)	123.1(2)
C(18)-C(20)	1.536(3)	C(5)-C(6)-C(7)	117.8(2)
C(18)-C(19)	1.538(3)	N(1)-C(7)-C(8)	120.8(2)
C(18)-C(21)	1.541(3)	N(1)-C(7)-C(6)	118.1(2)
C(22)-C(23)	1.517(3)	C(8)-C(7)-C(6)	121.0(2)
C(22)-C(25)	1.524(3)	C(9)-C(8)-C(7)	119.6(2)
C(22)-C(24)	1.534(3)	C(8)-C(9)-C(10)	119.6(2)
C(26)-C(27)	1.522(3)	C(9)-C(10)-C(11)	118.8(2)
C(26)-C(28)	1.531(3)	N(1)-C(11)-C(10)	121.8(2)
C(26)-C(29)	1.533(3)	N(1)-C(11)-C(12)	118.16(19)
C(30)-C(32)	1.533(3)	C(10)-C(11)-C(12)	120.0(2)
C(30)-C(33)	1.538(3)	C(13)-C(12)-C(17)	119.5(2)
C(30)-C(31)	1.543(3)	C(13)-C(12)-C(11)	118.5(2)
C(41)-C(42)	1.464(4)	C(17)-C(12)-C(11)	122.0(2)
C(42)-C(43)	1.397(5)	C(14)-C(13)-C(12)	122.0(2)
C(43)-C(44)	1.515(6)	C(15)-C(14)-C(13)	116.8(2)
C(44)-C(45)	1.415(6)	C(15)-C(14)-C(26)	123.2(2)

C(13)-C(14)-C(26)	120.0(2)
C(14)-C(15)-C(16)	124.3(2)
C(15)-C(16)-C(17)	117.4(2)
C(15)-C(16)-C(30)	121.5(2)
C(17)-C(16)-C(30)	121.1(2)
O(2)-C(17)-C(12)	120.0(2)
O(2)-C(17)-C(16)	119.9(2)
C(12)-C(17)-C(16)	120.0(2)
C(20)-C(18)-C(19)	110.4(2)
C(20)-C(18)-C(2)	110.03(18)
C(19)-C(18)-C(2)	109.75(18)
C(20)-C(18)-C(21)	107.20(19)
C(19)-C(18)-C(21)	107.55(18)
C(2)-C(18)-C(21)	111.8(2)
C(23)-C(22)-C(25)	108.1(2)
C(23)-C(22)-C(4)	111.4(2)
C(25)-C(22)-C(4)	111.23(19)
C(23)-C(22)-C(24)	108.5(2)
C(25)-C(22)-C(24)	108.3(2)
C(4)-C(22)-C(24)	109.20(18)
C(27)-C(26)-C(28)	107.7(2)
C(27)-C(26)-C(29)	108.8(2)
C(28)-C(26)-C(29)	108.5(2)
C(27)-C(26)-C(14)	110.00(18)
C(28)-C(26)-C(14)	112.4(2)
C(29)-C(26)-C(14)	109.44(19)
C(32)-C(30)-C(33)	107.5(2)
C(32)-C(30)-C(16)	112.2(2)
C(33)-C(30)-C(16)	108.7(2)
C(32)-C(30)-C(31)	107.1(2)
C(33)-C(30)-C(31)	110.3(2)
C(16)-C(30)-C(31)	110.99(18)
C(43)-C(42)-C(41)	121.8(4)
C(42)-C(43)-C(44)	120.1(5)
C(45)-C(44)-C(43)	114.3(4)

Table 5. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^4$) for IAT26 (CCDC 754732). The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
Ti(1)	175(2)	140(2)	207(2)	-11(2)	26(2)	5(2)
O(1)	205(10)	208(9)	232(10)	43(7)	22(8)	-46(7)
O(2)	202(10)	253(9)	197(10)	-48(7)	8(8)	40(7)
N(1)	191(11)	92(10)	185(11)	8(8)	26(9)	9(8)
N(2)	207(12)	120(10)	256(13)	-22(9)	21(10)	0(9)
N(3)	211(12)	183(11)	266(12)	-20(10)	2(10)	12(10)
C(1)	203(14)	74(12)	214(14)	1(10)	42(11)	44(10)
C(2)	180(14)	89(12)	288(15)	-32(10)	84(12)	36(10)
C(3)	245(15)	74(12)	294(16)	3(11)	132(13)	5(10)
C(4)	250(15)	91(12)	222(14)	-30(10)	89(12)	33(10)
C(5)	213(14)	117(13)	246(15)	-52(10)	25(12)	11(10)
C(6)	161(14)	99(12)	199(14)	-21(10)	42(11)	13(10)
C(7)	154(13)	113(12)	219(14)	19(11)	5(11)	50(10)
C(8)	242(15)	132(12)	190(14)	-34(11)	14(12)	-2(11)
C(9)	235(15)	114(12)	264(16)	-15(11)	-12(12)	-30(10)
C(10)	165(14)	129(12)	266(15)	26(11)	24(12)	-14(10)
C(11)	130(12)	97(11)	193(13)	47(10)	5(10)	25(10)
C(12)	190(14)	49(11)	203(14)	38(10)	30(11)	-36(10)
C(13)	215(14)	85(13)	192(14)	48(10)	5(11)	-39(10)
C(14)	185(14)	96(12)	243(15)	48(10)	48(12)	6(10)
C(15)	231(15)	102(12)	220(14)	-7(10)	52(12)	16(10)
C(16)	218(14)	122(12)	194(14)	22(10)	2(12)	31(10)
C(17)	161(13)	82(12)	247(14)	43(10)	6(12)	25(10)
C(18)	126(13)	173(13)	304(16)	16(11)	58(12)	-1(10)
C(19)	236(15)	268(15)	330(17)	-44(12)	14(13)	-78(12)
C(20)	200(15)	315(16)	431(18)	59(13)	-2(14)	13(11)
C(21)	197(15)	296(15)	382(18)	-16(13)	47(14)	-25(11)
C(22)	307(16)	143(13)	226(15)	-29(11)	62(13)	3(11)
C(23)	490(20)	620(20)	282(18)	40(15)	152(16)	-25(16)
C(24)	670(20)	278(16)	257(17)	1(13)	57(16)	115(15)
C(25)	690(20)	317(16)	203(16)	-31(12)	8(15)	-111(14)
C(26)	192(14)	162(13)	267(15)	-1(11)	50(12)	17(11)
C(27)	244(17)	363(17)	740(20)	-57(15)	106(16)	-36(13)
C(28)	279(18)	620(20)	470(20)	-113(16)	107(16)	90(15)
C(29)	274(17)	471(18)	510(20)	111(16)	61(16)	107(14)
C(30)	261(16)	275(15)	247(16)	-82(12)	-30(13)	87(12)
C(31)	432(19)	485(17)	266(17)	-82(14)	-106(15)	239(15)
C(32)	348(18)	428(17)	274(17)	-127(13)	-38(14)	149(13)
C(33)	265(17)	416(17)	430(20)	-229(14)	-34(15)	15(13)
C(34)	660(20)	281(16)	376(19)	-76(13)	142(17)	90(14)
C(35)	374(19)	237(16)	650(20)	58(14)	107(17)	-28(13)
C(36)	354(18)	352(16)	432(19)	-19(14)	-139(15)	75(13)
C(37)	277(16)	360(16)	433(18)	-71(14)	-29(14)	28(13)
C(41)	1290(40)	630(20)	730(30)	80(20)	70(30)	220(20)
C(42)	1150(40)	610(30)	860(30)	20(20)	110(30)	30(20)
C(43)	1620(50)	1520(50)	1200(50)	760(40)	-740(40)	-760(40)

C(44)	850(40)	3320(80)	1560(60)	1700(60)	630(40)	1130(50)
C(45)	3060(80)	400(20)	600(30)	-130(20)	290(40)	-200(30)

CALIFORNIA INSTITUTE OF TECHNOLOGY
BECKMAN INSTITUTE
X-RAY CRYSTALLOGRAPHY LABORATORY



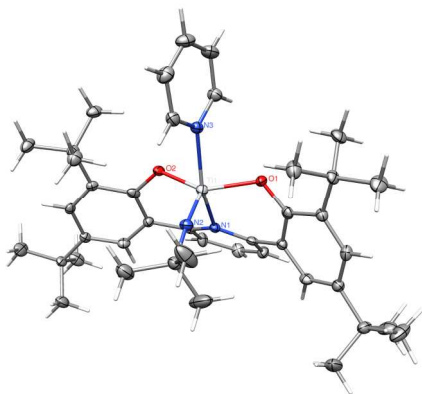
Date 5 October 2011

Crystal Structure Analysis of:
Complex 4 (ONO)Ti(py)(N^tBu) (IAT50)
(shown below)

By Michael W. Day 116 Beckman ext. 2734
e-mail: mikeday@caltech.edu

Contents

Table 1. Crystal data
Figures Minimum overlap
Table 2. Atomic Coordinates
Table 3. Selected bond distances and angles
Table 4. Full bond distances and angles
Table 5. Anisotropic displacement parameters
Table 6. Observed and calculated structure factors (available upon request)



IAT50

Note: The crystallographic data have been deposited in the Cambridge Database (CCDC). The deposition number is 847140.

Table 1. Crystal data and structure refinement for IAT50 (CCDC 847140).

Empirical formula	$\text{C}_{42}\text{H}_{57}\text{N}_3\text{O}_2\text{Ti} \cdot 0.49(\text{C}_5\text{H}_{12}) \cdot 0.51(\text{C}_7\text{H}_8)$		
Formula weight	765.95		
Crystallization Solvent	Toluene/pentane		
Crystal Habit	Block		
Crystal size	0.17 x 0.14 x 0.08 mm ³		
Crystal color	Yellow		
Data Collection			
Type of diffractometer	Bruker KAPPA APEX II		
Wavelength	0.71073 Å MoKα		
Data Collection Temperature	100(2) K		
θ range for 9969 reflections used in lattice determination	2.16 to 28.75°		
Unit cell dimensions	a = 14.6930(8) Å b = 14.9636(8) Å c = 20.1744(12) Å	α= 90° β= 92.699(3)° γ = 90°	
Volume	4430.6(4) Å ³		
Z	4		
Crystal system	Monoclinic		
Space group	P 2 ₁ /c		
Density (calculated)	1.148 Mg/m ³		
F(000)	1656		
θ range for data collection	1.70 to 30.51°		
Completeness to θ = 27.50°	100.0 %		
Index ranges	-20 ≤ h ≤ 20, -21 ≤ k ≤ 20, -28 ≤ l ≤ 27		
Data collection scan type	ω scans; 8 settings		
Reflections collected	72499		
Independent reflections	12470 [R _{int} = 0.0714]		
Absorption coefficient	0.233 mm ⁻¹		
Absorption correction	None		
Max. and min. transmission	0.9816 and 0.9615		



Table 1 (cont.)**Structure solution and Refinement**

Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on F^2
Data / restraints / parameters	12470 / 8 / 485
Treatment of hydrogen atoms	Riding
Goodness-of-fit on F^2	1.945
Final R indices [$I > 2\sigma(I)$, 7342 reflections]	$R1 = 0.0584$, $wR2 = 0.0723$
R indices (all data)	$R1 = 0.1107$, $wR2 = 0.0742$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.001
Average shift/error	0.000
Largest diff. peak and hole	1.021 and -0.754 e.Å ⁻³

Special Refinement Details

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

The solvent region of this crystal is combination of toluene and pentane in nearly equal amounts, 51:49 respectively. The toluene was restrained to be flat with the six member ring constrained to a regular hexagon and the pentane was restrained to have similar carbon-carbon bond distances. Both were refined isotropically.

Refinement of F^2 against ALL reflections. The weighted R-factor (wR) and goodness of fit (S) are based on F^2 , conventional R-factors (R) are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

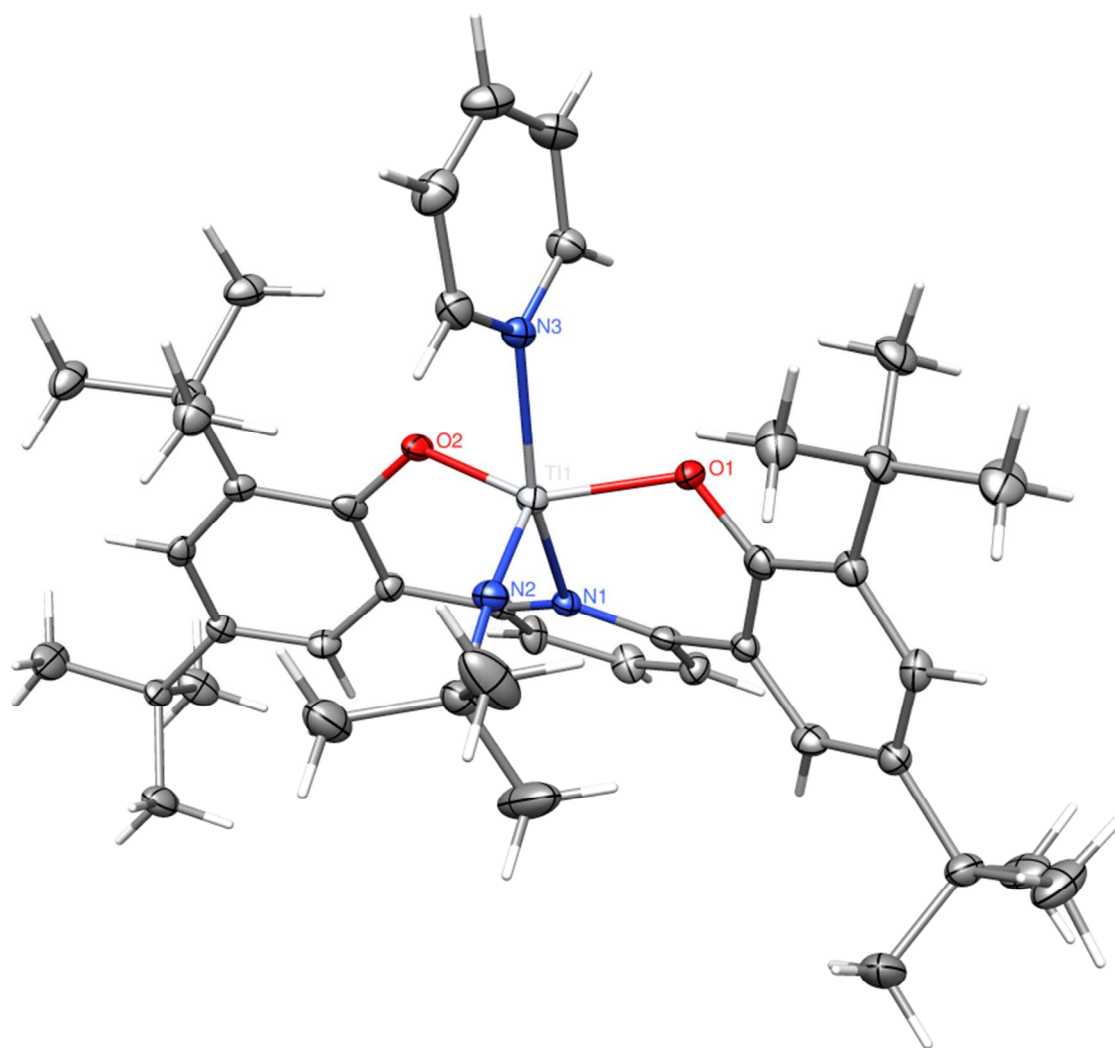


Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for IAT50 (CCDC 847140). $U(\text{eq})$ is defined as the trace of the orthogonalized U^i tensor.

	x	y	z	U_{eq}	Occ
Ti(1)	3121(1)	4119(1)	8795(1)	17(1)	1
O(1)	2405(1)	5035(1)	9187(1)	17(1)	1
O(2)	4090(1)	3341(1)	9121(1)	17(1)	1
N(1)	2226(1)	3180(1)	9314(1)	15(1)	1
N(2)	2779(1)	3924(1)	7997(1)	18(1)	1
N(3)	4190(1)	5152(1)	8722(1)	17(1)	1
C(1)	1492(1)	5053(2)	9058(1)	18(1)	1
C(2)	1043(1)	5823(2)	8798(1)	17(1)	1
C(3)	114(1)	5751(2)	8636(1)	20(1)	1
C(4)	-397(1)	4985(2)	8735(1)	18(1)	1
C(5)	48(1)	4266(1)	9036(1)	20(1)	1
C(6)	982(1)	4287(1)	9194(1)	15(1)	1
C(7)	1408(1)	3488(1)	9518(1)	17(1)	1
C(8)	950(1)	3052(2)	10012(1)	21(1)	1
C(9)	1320(1)	2299(1)	10304(1)	24(1)	1
C(10)	2120(1)	1959(1)	10077(1)	22(1)	1
C(11)	2556(1)	2391(1)	9568(1)	17(1)	1
C(12)	3354(1)	1954(1)	9283(1)	16(1)	1
C(13)	3360(1)	1026(1)	9233(1)	19(1)	1
C(14)	4088(1)	555(1)	8994(1)	16(1)	1
C(15)	4826(1)	1062(1)	8800(1)	16(1)	1
C(16)	4868(1)	1984(1)	8832(1)	15(1)	1
C(17)	4104(1)	2447(1)	9071(1)	17(1)	1
C(18)	1568(1)	6697(1)	8687(1)	19(1)	1
C(19)	2250(1)	6566(1)	8140(1)	27(1)	1
C(20)	932(1)	7471(1)	8467(1)	29(1)	1
C(21)	2073(1)	6983(1)	9338(1)	27(1)	1
C(22)	-1423(1)	4911(2)	8534(1)	22(1)	1
C(23)	-1776(1)	5746(2)	8168(1)	40(1)	1
C(24)	-1963(1)	4801(2)	9158(1)	33(1)	1
C(25)	-1582(1)	4099(2)	8078(1)	41(1)	1
C(26)	4048(1)	-466(1)	8941(1)	18(1)	1
C(27)	4942(1)	-869(2)	8716(1)	32(1)	1
C(28)	3284(1)	-724(1)	8429(1)	27(1)	1
C(29)	3856(1)	-875(2)	9617(1)	30(1)	1
C(30)	5709(1)	2498(1)	8616(1)	18(1)	1
C(31)	6494(1)	1867(1)	8441(1)	27(1)	1
C(32)	6079(1)	3119(1)	9176(1)	26(1)	1
C(33)	5455(1)	3048(1)	7990(1)	25(1)	1
C(34)	2367(1)	3674(2)	7357(1)	22(1)	1
C(35)	2378(2)	4477(2)	6889(1)	43(1)	1
C(36)	2914(1)	2914(2)	7063(1)	34(1)	1
C(37)	1384(1)	3380(2)	7449(1)	53(1)	1
C(38)	4620(1)	5336(1)	8167(1)	21(1)	1
C(39)	5345(1)	5919(2)	8147(1)	27(1)	1
C(40)	5638(2)	6333(2)	8731(1)	30(1)	1
C(41)	5200(1)	6166(2)	9301(1)	28(1)	1
C(42)	4475(1)	5576(1)	9277(1)	22(1)	1

C(51A)	8507(3)	709(3)	9692(2)	41(2)	0.507(3)
C(52A)	9238(3)	807(4)	9245(2)	140(4)	0.507(3)
C(53A)	9886(4)	144(3)	9167(3)	127(3)	0.507(3)
C(54A)	10615(4)	296(3)	8767(3)	128(4)	0.507(3)
C(55A)	10695(3)	1111(4)	8444(2)	223(7)	0.507(3)
C(56A)	10047(4)	1775(3)	8523(2)	85(3)	0.507(3)
C(57A)	9318(3)	1623(3)	8923(3)	95(3)	0.507(3)
C(51B)	10844(3)	920(5)	8703(3)	65(2)	0.493(3)
C(52B)	9898(3)	827(5)	8353(3)	96(3)	0.493(3)
C(53B)	9208(4)	1477(4)	8626(3)	75(2)	0.493(3)
C(54B)	9091(6)	1375(7)	9389(4)	145(4)	0.493(3)
C(55B)	8925(10)	791(11)	10026(7)	485(14)	0.493(3)

Table 3. Selected bond lengths [Å] and angles [°] for IAT50 (CCDC 847140).

Ti(1)-N(2)	1.6903(16)	N(2)-Ti(1)-O(1)	111.81(7)
Ti(1)-O(1)	1.9204(14)	N(2)-Ti(1)-O(2)	113.70(7)
Ti(1)-O(2)	1.9309(13)	O(1)-Ti(1)-O(2)	134.39(6)
Ti(1)-N(3)	2.2132(17)	N(2)-Ti(1)-N(3)	103.72(7)
Ti(1)-N(1)	2.2205(17)	O(1)-Ti(1)-N(3)	86.21(6)
		O(2)-Ti(1)-N(3)	85.98(6)
		N(2)-Ti(1)-N(1)	100.56(7)
		O(1)-Ti(1)-N(1)	84.80(6)
		O(2)-Ti(1)-N(1)	84.33(6)
		N(3)-Ti(1)-N(1)	155.72(6)

Table 4. Bond lengths [Å] and angles [°] for IAT50 (CCDC 847140).

Ti(1)-N(2)	1.6903(16)	C(40)-C(41)	1.368(3)
Ti(1)-O(1)	1.9204(14)	C(41)-C(42)	1.382(3)
Ti(1)-O(2)	1.9309(13)	C(51A)-C(52A)	1.441(6)
Ti(1)-N(3)	2.2132(17)	C(52A)-C(53A)	1.3900
Ti(1)-N(1)	2.2205(17)	C(52A)-C(57A)	1.3900
O(1)-C(1)	1.354(2)	C(53A)-C(54A)	1.3900
O(2)-C(17)	1.342(2)	C(54A)-C(55A)	1.3900
N(1)-C(11)	1.366(2)	C(55A)-C(56A)	1.3900
N(1)-C(7)	1.368(2)	C(56A)-C(57A)	1.3900
N(2)-C(34)	1.448(2)	C(51B)-C(52B)	1.535(4)
N(3)-C(42)	1.338(2)	C(52B)-C(53B)	1.525(4)
N(3)-C(38)	1.339(2)	C(53B)-C(54B)	1.565(5)
C(1)-C(6)	1.402(3)	C(54B)-C(55B)	1.582(5)
C(1)-C(2)	1.417(3)		
C(2)-C(3)	1.392(2)	N(2)-Ti(1)-O(1)	111.81(7)
C(2)-C(18)	1.541(3)	N(2)-Ti(1)-O(2)	113.70(7)
C(3)-C(4)	1.390(3)	O(1)-Ti(1)-O(2)	134.39(6)
C(4)-C(5)	1.384(3)	N(2)-Ti(1)-N(3)	103.72(7)
C(4)-C(22)	1.546(3)	O(1)-Ti(1)-N(3)	86.21(6)
C(5)-C(6)	1.395(2)	O(2)-Ti(1)-N(3)	85.98(6)
C(6)-C(7)	1.487(3)	N(2)-Ti(1)-N(1)	100.56(7)
C(7)-C(8)	1.391(3)	O(1)-Ti(1)-N(1)	84.80(6)
C(8)-C(9)	1.371(3)	O(2)-Ti(1)-N(1)	84.33(6)
C(9)-C(10)	1.380(3)	N(3)-Ti(1)-N(1)	155.72(6)
C(10)-C(11)	1.395(3)	C(1)-O(1)-Ti(1)	119.47(12)
C(11)-C(12)	1.482(3)	C(17)-O(2)-Ti(1)	126.09(12)
C(12)-C(13)	1.393(3)	C(11)-N(1)-C(7)	118.63(18)
C(12)-C(17)	1.409(3)	C(11)-N(1)-Ti(1)	121.08(14)
C(13)-C(14)	1.386(3)	C(7)-N(1)-Ti(1)	118.47(14)
C(14)-C(15)	1.394(3)	C(34)-N(2)-Ti(1)	170.76(15)
C(14)-C(26)	1.533(3)	C(42)-N(3)-C(38)	117.45(19)
C(15)-C(16)	1.383(3)	C(42)-N(3)-Ti(1)	118.05(15)
C(16)-C(17)	1.422(3)	C(38)-N(3)-Ti(1)	124.31(15)
C(16)-C(30)	1.535(3)	O(1)-C(1)-C(6)	118.66(19)
C(18)-C(19)	1.537(3)	O(1)-C(1)-C(2)	121.80(19)
C(18)-C(21)	1.538(3)	C(6)-C(1)-C(2)	119.53(19)
C(18)-C(20)	1.540(3)	C(3)-C(2)-C(1)	117.3(2)
C(22)-C(23)	1.529(3)	C(3)-C(2)-C(18)	121.59(19)
C(22)-C(24)	1.530(3)	C(1)-C(2)-C(18)	121.13(17)
C(22)-C(25)	1.535(3)	C(4)-C(3)-C(2)	124.1(2)
C(26)-C(29)	1.533(3)	C(5)-C(4)-C(3)	117.19(19)
C(26)-C(27)	1.533(3)	C(5)-C(4)-C(22)	119.6(2)
C(26)-C(28)	1.538(2)	C(3)-C(4)-C(22)	123.21(19)
C(30)-C(33)	1.538(3)	C(4)-C(5)-C(6)	121.4(2)
C(30)-C(32)	1.542(3)	C(5)-C(6)-C(1)	120.2(2)
C(30)-C(31)	1.545(3)	C(5)-C(6)-C(7)	118.09(19)
C(34)-C(35)	1.528(3)	C(1)-C(6)-C(7)	121.68(18)
C(34)-C(36)	1.528(3)	N(1)-C(7)-C(8)	121.3(2)
C(34)-C(37)	1.529(3)	N(1)-C(7)-C(6)	119.70(19)
C(38)-C(39)	1.380(3)	C(8)-C(7)-C(6)	118.98(19)
C(39)-C(40)	1.380(3)	C(9)-C(8)-C(7)	119.8(2)

C(8)-C(9)-C(10)	119.2(2)	C(53A)-C(52A)-C(57A)	120.0
C(9)-C(10)-C(11)	120.0(2)	C(53A)-C(52A)-C(51A)	122.2(5)
N(1)-C(11)-C(10)	120.7(2)	C(57A)-C(52A)-C(51A)	117.7(5)
N(1)-C(11)-C(12)	120.6(2)	C(54A)-C(53A)-C(52A)	120.0
C(10)-C(11)-C(12)	118.6(2)	C(53A)-C(54A)-C(55A)	120.0
C(13)-C(12)-C(17)	119.5(2)	C(56A)-C(55A)-C(54A)	120.0
C(13)-C(12)-C(11)	118.42(19)	C(57A)-C(56A)-C(55A)	120.0
C(17)-C(12)-C(11)	122.1(2)	C(56A)-C(57A)-C(52A)	120.0
C(14)-C(13)-C(12)	122.6(2)	C(53B)-C(52B)-C(51B)	112.2(5)
C(13)-C(14)-C(15)	116.4(2)	C(52B)-C(53B)-C(54B)	113.5(6)
C(13)-C(14)-C(26)	120.20(19)	C(53B)-C(54B)-C(55B)	151.8(11)
C(15)-C(14)-C(26)	123.39(19)		
C(16)-C(15)-C(14)	124.3(2)		
C(15)-C(16)-C(17)	117.8(2)		
C(15)-C(16)-C(30)	121.43(19)		
C(17)-C(16)-C(30)	120.75(19)		
O(2)-C(17)-C(12)	118.9(2)		
O(2)-C(17)-C(16)	121.7(2)		
C(12)-C(17)-C(16)	119.3(2)		
C(19)-C(18)-C(21)	110.17(17)		
C(19)-C(18)-C(2)	110.10(17)		
C(21)-C(18)-C(2)	109.62(17)		
C(19)-C(18)-C(20)	107.26(18)		
C(21)-C(18)-C(20)	107.37(18)		
C(2)-C(18)-C(20)	112.26(17)		
C(23)-C(22)-C(24)	107.92(19)		
C(23)-C(22)-C(25)	108.62(19)		
C(24)-C(22)-C(25)	109.73(19)		
C(23)-C(22)-C(4)	111.90(18)		
C(24)-C(22)-C(4)	109.24(17)		
C(25)-C(22)-C(4)	109.40(18)		
C(29)-C(26)-C(27)	107.64(18)		
C(29)-C(26)-C(14)	110.08(18)		
C(27)-C(26)-C(14)	112.44(18)		
C(29)-C(26)-C(28)	109.78(18)		
C(27)-C(26)-C(28)	108.15(18)		
C(14)-C(26)-C(28)	108.72(17)		
C(16)-C(30)-C(33)	109.52(16)		
C(16)-C(30)-C(32)	110.76(18)		
C(33)-C(30)-C(32)	109.99(18)		
C(16)-C(30)-C(31)	112.27(18)		
C(33)-C(30)-C(31)	107.10(18)		
C(32)-C(30)-C(31)	107.11(17)		
N(2)-C(34)-C(35)	109.40(18)		
N(2)-C(34)-C(36)	109.47(17)		
C(35)-C(34)-C(36)	108.9(2)		
N(2)-C(34)-C(37)	108.85(18)		
C(35)-C(34)-C(37)	109.79(19)		
C(36)-C(34)-C(37)	110.44(19)		
N(3)-C(38)-C(39)	123.3(2)		
C(40)-C(39)-C(38)	118.0(2)		
C(41)-C(40)-C(39)	119.7(2)		
C(40)-C(41)-C(42)	118.6(2)		
N(3)-C(42)-C(41)	122.9(2)		

Table 5. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^4$) for IAT50 (CCDC 847140). The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
Ti(1)	166(2)	146(2)	196(2)	-5(2)	20(2)	-16(2)
O(1)	131(8)	152(9)	216(9)	-11(7)	9(7)	-8(7)
O(2)	157(8)	124(9)	229(10)	-11(7)	-6(7)	-20(7)
N(1)	152(10)	135(11)	168(11)	-13(9)	15(8)	-20(8)
N(2)	176(10)	190(12)	163(11)	-12(9)	8(8)	-33(8)
N(3)	165(10)	160(11)	176(11)	-2(9)	1(9)	-13(8)
C(1)	199(13)	178(14)	155(13)	-32(11)	42(10)	-1(11)
C(2)	182(12)	148(13)	185(13)	1(11)	45(10)	-9(11)
C(3)	204(13)	189(14)	204(14)	30(11)	47(10)	39(11)
C(4)	176(13)	184(14)	187(14)	-9(11)	15(10)	-5(11)
C(5)	195(13)	182(14)	213(14)	0(11)	41(10)	-63(11)
C(6)	156(12)	148(14)	148(13)	-10(10)	30(10)	-8(10)
C(7)	174(13)	149(13)	193(14)	-40(11)	-7(10)	-47(10)
C(8)	178(13)	205(14)	242(15)	-3(11)	50(11)	-11(11)
C(9)	302(15)	213(15)	208(15)	53(11)	102(11)	-21(12)
C(10)	268(14)	150(13)	245(15)	39(11)	72(11)	37(11)
C(11)	189(13)	137(13)	179(14)	-30(10)	-11(10)	-33(10)
C(12)	151(12)	162(13)	167(13)	2(11)	19(10)	-6(10)
C(13)	188(13)	174(14)	198(14)	28(11)	17(10)	-52(11)
C(14)	185(13)	151(13)	139(13)	16(10)	-13(10)	0(10)
C(15)	136(12)	188(14)	148(13)	8(10)	-13(9)	32(10)
C(16)	146(12)	164(13)	140(13)	24(10)	-11(10)	-19(10)
C(17)	205(13)	150(13)	146(13)	11(10)	-59(10)	-12(11)
C(18)	210(13)	109(13)	258(15)	24(11)	31(11)	2(10)
C(19)	285(14)	200(15)	333(17)	41(12)	122(12)	0(11)
C(20)	283(15)	169(14)	428(18)	85(12)	62(12)	-40(11)
C(21)	310(15)	143(14)	355(17)	-23(12)	34(12)	-40(11)
C(22)	185(13)	201(14)	258(15)	17(12)	-45(11)	-12(11)
C(23)	193(14)	387(19)	600(20)	149(15)	-47(13)	-34(13)
C(24)	164(13)	447(18)	390(18)	-16(14)	26(12)	-104(12)
C(25)	335(15)	405(18)	473(19)	-189(16)	-176(13)	51(14)
C(26)	178(13)	127(13)	238(15)	-21(11)	12(11)	-7(10)
C(27)	303(14)	169(14)	494(18)	-23(14)	33(12)	1(12)
C(28)	285(14)	224(16)	305(16)	-70(12)	-34(12)	-16(11)
C(29)	429(15)	199(14)	279(15)	33(13)	-23(12)	-16(13)
C(30)	166(13)	170(13)	204(14)	20(11)	20(11)	-22(10)
C(31)	195(13)	227(15)	389(17)	42(12)	70(12)	-19(11)
C(32)	216(13)	232(15)	336(16)	26(12)	8(12)	-74(11)
C(33)	242(14)	234(15)	289(16)	53(12)	55(11)	-6(11)
C(34)	220(14)	229(15)	197(15)	-47(11)	10(11)	-24(11)
C(35)	730(20)	283(17)	267(17)	-23(13)	-125(15)	87(14)
C(36)	391(16)	291(17)	323(17)	-83(13)	-41(13)	-7(13)
C(37)	271(16)	940(30)	390(20)	-261(17)	15(13)	-187(16)
C(38)	236(14)	192(14)	209(15)	-4(11)	21(11)	2(11)
C(39)	243(13)	285(15)	297(16)	64(13)	76(11)	-34(13)
C(40)	247(14)	277(16)	371(18)	31(13)	-4(13)	-104(12)
C(41)	291(15)	266(16)	290(16)	-34(12)	-38(12)	-69(12)

C(42)	233(14)	223(14)	200(15)	1(11)	36(11)	-40(11)
-------	---------	---------	---------	-------	--------	---------

CALIFORNIA INSTITUTE OF TECHNOLOGY
BECKMAN INSTITUTE
X-RAY CRYSTALLOGRAPHY LABORATORY



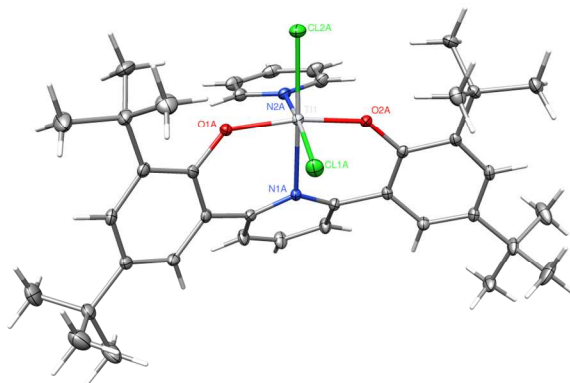
Date 2 December 2011

Crystal Structure Analysis of:
Complex 6 (ONO)TiCl₂(py) (IAT98)
(shown below)

By Michael W. Day 116 Beckman ext. 2734
e-mail: mikeday@caltech.edu

Contents

Table 1. Crystal data
Figures Minimum overlap
Table 2. Atomic Coordinates
Table 3. Selected bond distances and angles
Table 4. Full bond distances and angles
Table 5. Anisotropic displacement parameters
Table 6. Observed and calculated structure factors (available upon request)



IAT98

Note: The crystallographic data have been deposited in the Cambridge Database (CCDC). The deposition number is 856446.

Table 1. Crystal data and structure refinement for IAT98 (CCDC 856446).

Empirical formula	$2(\text{C}_{38}\text{H}_{48}\text{N}_2\text{O}_2\text{Cl}_2\text{Ti}) \cdot 5(\text{C}_6\text{H}_6)$
Formula weight	878.85
Crystallization Solvent	Dichloromethane
Crystal Habit	Block
Crystal size	0.36 x 0.25 x 0.20 mm ³
Crystal color	Orange-red



Data Collection

Type of diffractometer	Bruker KAPPA APEX II
Wavelength	0.71073 Å MoK α
Data Collection Temperature	100(2) K
θ range for 9190 reflections used in lattice determination	2.55 to 34.09°
Unit cell dimensions	$a = 11.8721(6)$ Å $b = 28.5845(14)$ Å $c = 29.6698(15)$ Å
	$\alpha = 90^\circ$ $\beta = 100.018(3)^\circ$ $\gamma = 90^\circ$
Volume	9915.2(9) Å ³
Z	8
Crystal system	Monoclinic
Space group	P 2 ₁ /c
Density (calculated)	1.177 Mg/m ³
F(000)	3736
Data collection program	Bruker APEX2 v2009.7-0
θ range for data collection	1.59 to 35.13°
Completeness to $\theta = 35.13^\circ$	99.7 %
Index ranges	$-13 \leq h \leq 19$, $-46 \leq k \leq 46$, $-48 \leq l \leq 48$
Data collection scan type	ω scans; 12 settings
Data reduction program	Bruker SAINT-Plus v7.68A
Reflections collected	325048
Independent reflections	43933 [$R_{\text{int}} = 0.0602$]
Absorption coefficient	0.320 mm ⁻¹
Absorption correction	None
Max. and min. transmission	0.9387 and 0.8934

Table 1 (cont.)**Structure solution and Refinement**

Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on F^2
Data / restraints / parameters	43933 / 0 / 1105
Treatment of hydrogen atoms	Riding
Goodness-of-fit on F^2	1.877
Final R indices [$I > 2\sigma(I)$, 30745 reflections]	$R1 = 0.0415$, $wR2 = 0.0625$
R indices (all data)	$R1 = 0.0689$, $wR2 = 0.0636$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.004
Average shift/error	0.000
Largest diff. peak and hole	0.756 and -0.589 e.Å ⁻³

Special Refinement Details

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

Refinement of F^2 against ALL reflections. The weighted R-factor (wR) and goodness of fit (S) are based on F^2 , conventional R-factors (R) are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

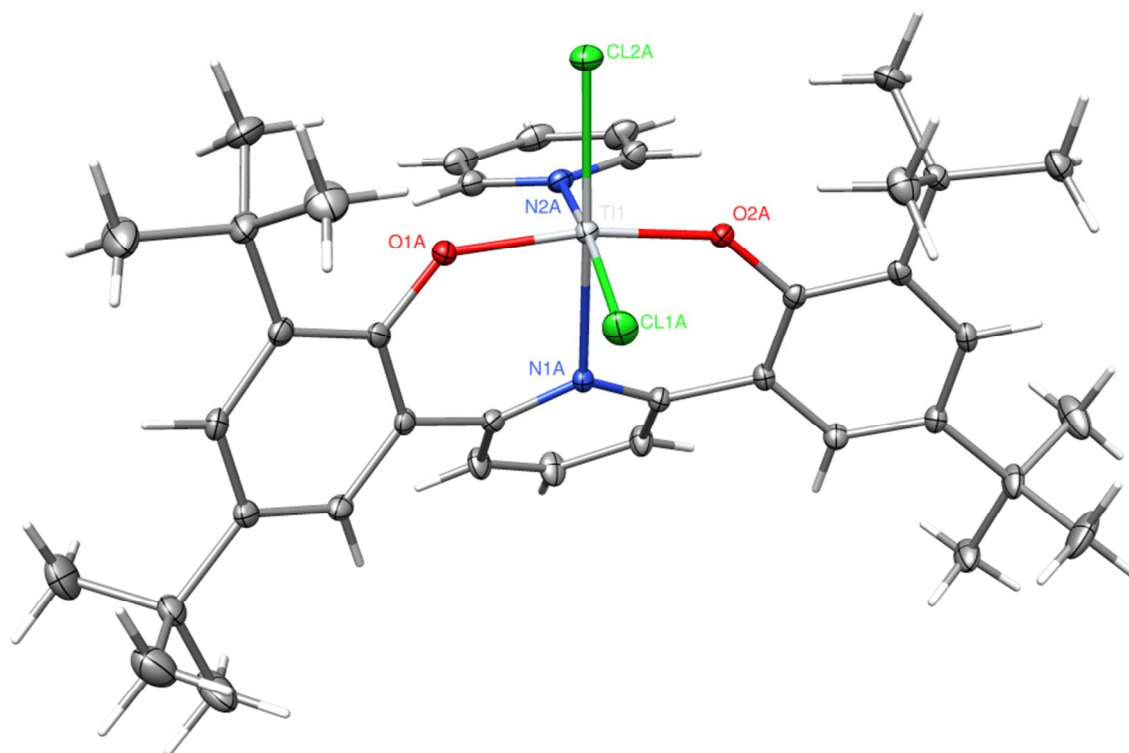


Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for IAT98 (CCDC 856446). $U(\text{eq})$ is defined as the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U_{eq}
Ti(1)	5235(1)	1029(1)	6098(1)	11(1)
Cl(1A)	5402(1)	1419(1)	6783(1)	19(1)
Cl(2A)	3740(1)	536(1)	6133(1)	18(1)
O(1A)	6375(1)	616(1)	6336(1)	13(1)
O(2A)	4383(1)	1498(1)	5803(1)	12(1)
N(1A)	6724(1)	1463(1)	5921(1)	11(1)
N(2A)	5375(1)	755(1)	5398(1)	14(1)
C(1A)	7462(1)	641(1)	6559(1)	12(1)
C(2A)	7874(1)	333(1)	6922(1)	14(1)
C(3A)	9007(1)	388(1)	7129(1)	15(1)
C(4A)	9748(1)	725(1)	6999(1)	14(1)
C(5A)	9310(1)	1012(1)	6636(1)	14(1)
C(6A)	8178(1)	973(1)	6409(1)	12(1)
C(7A)	7794(1)	1272(1)	6002(1)	13(1)
C(8A)	8578(1)	1359(1)	5712(1)	17(1)
C(9A)	8298(1)	1657(1)	5346(1)	19(1)
C(10A)	7269(1)	1892(1)	5294(1)	17(1)
C(11A)	6504(1)	1803(1)	5591(1)	13(1)
C(12A)	5461(1)	2101(1)	5550(1)	12(1)
C(13A)	5513(1)	2563(1)	5396(1)	14(1)
C(14A)	4567(1)	2854(1)	5341(1)	14(1)
C(15A)	3548(1)	2672(1)	5446(1)	13(1)
C(16A)	3442(1)	2223(1)	5607(1)	11(1)
C(17A)	4431(1)	1941(1)	5658(1)	11(1)
C(18A)	7099(1)	-44(1)	7078(1)	20(1)
C(19A)	6126(1)	186(1)	7278(1)	27(1)
C(20A)	6599(1)	-356(1)	6673(1)	28(1)
C(21A)	7774(1)	-360(1)	7448(1)	35(1)
C(22A)	10974(1)	764(1)	7260(1)	19(1)
C(23A)	11590(1)	295(1)	7247(1)	36(1)
C(24A)	10950(1)	888(1)	7766(1)	33(1)
C(25A)	11650(1)	1141(1)	7064(1)	39(1)
C(26A)	4590(1)	3361(1)	5176(1)	19(1)
C(27A)	5749(1)	3494(1)	5052(1)	24(1)
C(28A)	4357(1)	3691(1)	5557(1)	35(1)
C(29A)	3661(1)	3428(1)	4754(1)	32(1)
C(30A)	2320(1)	2036(1)	5732(1)	13(1)
C(31A)	1913(1)	1598(1)	5450(1)	20(1)
C(32A)	1368(1)	2402(1)	5634(1)	19(1)
C(33A)	2492(1)	1921(1)	6244(1)	19(1)
C(34A)	6160(1)	432(1)	5336(1)	17(1)
C(35A)	6303(1)	278(1)	4909(1)	22(1)
C(36A)	5632(1)	463(1)	4526(1)	23(1)
C(37A)	4821(1)	796(1)	4584(1)	22(1)
C(38A)	4718(1)	929(1)	5020(1)	18(1)

Ti(2)	8749(1)	2409(1)	8959(1)	11(1)
Cl(1B)	7058(1)	2437(1)	9247(1)	17(1)
Cl(2B)	9902(1)	2874(1)	9468(1)	17(1)
O(1B)	8272(1)	2876(1)	8549(1)	13(1)
O(2B)	9136(1)	1854(1)	9261(1)	14(1)
N(1B)	7883(1)	1938(1)	8394(1)	12(1)
N(2B)	10215(1)	2290(1)	8589(1)	15(1)
C(1B)	7429(1)	2966(1)	8194(1)	12(1)
C(2B)	6963(1)	3418(1)	8119(1)	14(1)
C(3B)	6067(1)	3467(1)	7751(1)	18(1)
C(4B)	5631(1)	3101(1)	7461(1)	19(1)
C(5B)	6153(1)	2667(1)	7540(1)	17(1)
C(6B)	7053(1)	2594(1)	7901(1)	13(1)
C(7B)	7562(1)	2119(1)	7964(1)	13(1)
C(8B)	7614(1)	1854(1)	7576(1)	16(1)
C(9B)	7921(1)	1389(1)	7620(1)	19(1)
C(10B)	8092(1)	1191(1)	8047(1)	18(1)
C(11B)	8043(1)	1465(1)	8431(1)	14(1)
C(12B)	8102(1)	1231(1)	8880(1)	14(1)
C(13B)	7564(1)	794(1)	8891(1)	17(1)
C(14B)	7595(1)	556(1)	9298(1)	18(1)
C(15B)	8177(1)	764(1)	9697(1)	18(1)
C(16B)	8719(1)	1197(1)	9709(1)	15(1)
C(17B)	8647(1)	1430(1)	9285(1)	14(1)
C(18B)	7430(1)	3830(1)	8427(1)	18(1)
C(19B)	6823(1)	4290(1)	8268(1)	29(1)
C(20B)	8712(1)	3898(1)	8418(1)	24(1)
C(21B)	7249(1)	3741(1)	8920(1)	23(1)
C(22B)	4594(1)	3151(1)	7078(1)	31(1)
C(23B)	4220(2)	3657(1)	6999(1)	64(1)
C(24B)	4862(1)	2952(1)	6633(1)	34(1)
C(25B)	3589(1)	2869(1)	7212(1)	48(1)
C(26B)	7014(1)	80(1)	9330(1)	21(1)
C(27B)	6443(1)	-103(1)	8862(1)	29(1)
C(28B)	6108(1)	125(1)	9637(1)	28(1)
C(29B)	7928(1)	-279(1)	9542(1)	28(1)
C(30B)	9389(1)	1399(1)	10153(1)	17(1)
C(31B)	10653(1)	1444(1)	10102(1)	21(1)
C(32B)	8926(1)	1880(1)	10263(1)	20(1)
C(33B)	9332(1)	1079(1)	10564(1)	26(1)
C(34B)	10306(1)	2515(1)	8202(1)	23(1)
C(35B)	11205(1)	2447(1)	7970(1)	33(1)
C(36B)	12058(1)	2140(1)	8145(1)	32(1)
C(37B)	11980(1)	1908(1)	8541(1)	32(1)
C(38B)	11055(1)	1989(1)	8753(1)	25(1)
C(41)	8180(2)	4688(1)	6414(1)	64(1)
C(42)	9339(2)	4707(1)	6516(1)	52(1)
C(43)	9924(2)	5037(1)	6319(1)	46(1)
C(44)	9370(2)	5354(1)	6018(1)	49(1)
C(45)	8210(2)	5344(1)	5908(1)	63(1)
C(46)	7588(2)	5009(1)	6104(1)	75(1)

C(51)	8347(2)	2859(1)	6771(1)	111(1)
C(52)	9269(2)	2716(1)	6576(1)	111(1)
C(53)	10277(2)	2944(1)	6678(1)	117(1)
C(54)	10416(2)	3321(1)	6978(1)	106(1)
C(55)	9501(2)	3470(1)	7175(1)	107(1)
C(56)	8486(3)	3234(1)	7072(1)	125(1)
C(61)	4416(1)	3057(1)	8772(1)	33(1)
C(62)	3689(1)	3337(1)	8483(1)	32(1)
C(63)	2625(1)	3445(1)	8580(1)	35(1)
C(64)	2294(1)	3272(1)	8967(1)	37(1)
C(65)	3031(1)	2992(1)	9258(1)	39(1)
C(66)	4093(1)	2886(1)	9160(1)	37(1)
C(71)	5116(1)	1743(1)	8330(1)	32(1)
C(72)	4951(1)	1501(1)	8710(1)	34(1)
C(73)	4545(1)	1049(1)	8670(1)	38(1)
C(74)	4326(1)	836(1)	8245(1)	34(1)
C(75)	4518(1)	1073(1)	7864(1)	28(1)
C(76)	4898(1)	1530(1)	7904(1)	32(1)
C(81)	420(2)	4310(1)	9620(1)	49(1)
C(82)	1120(1)	4507(1)	9324(1)	51(1)
C(83)	737(1)	4898(1)	9077(1)	42(1)
C(84)	-278(1)	5090(1)	9115(1)	36(1)
C(85)	-937(1)	4906(1)	9394(1)	34(1)
C(86)	-605(1)	4522(1)	9646(1)	35(1)

Table 3. Selected bond lengths [\AA] and angles [$^\circ$] for IAT98 (CCDC 856446).

Ti(1)-O(2A)	1.8119(7)	Ti(2)-O(1B)	1.8294(7)
Ti(1)-O(1A)	1.8426(7)	Ti(2)-O(2B)	1.8391(7)
Ti(1)-N(2A)	2.2533(9)	Ti(2)-N(2B)	2.2400(9)
Ti(1)-Cl(2A)	2.2832(3)	Ti(2)-N(1B)	2.2536(9)
Ti(1)-N(1A)	2.2950(8)	Ti(2)-Cl(2B)	2.2798(3)
Ti(1)-Cl(1A)	2.2961(3)	Ti(2)-Cl(1B)	2.3149(3)
<hr/>			
O(2A)-Ti(1)-O(1A)	166.32(3)	O(1B)-Ti(2)-O(2B)	167.35(3)
O(2A)-Ti(1)-N(2A)	86.38(3)	O(1B)-Ti(2)-N(2B)	87.79(3)
O(1A)-Ti(1)-N(2A)	87.98(3)	O(2B)-Ti(2)-N(2B)	87.65(3)
O(2A)-Ti(1)-Cl(2A)	96.07(2)	O(1B)-Ti(2)-N(1B)	83.83(3)
O(1A)-Ti(1)-Cl(2A)	96.43(2)	O(2B)-Ti(2)-N(1B)	83.80(3)
N(2A)-Ti(1)-Cl(2A)	90.59(2)	N(2B)-Ti(2)-N(1B)	80.64(3)
O(2A)-Ti(1)-N(1A)	82.89(3)	O(1B)-Ti(2)-Cl(2B)	95.44(2)
O(1A)-Ti(1)-N(1A)	83.75(3)	O(2B)-Ti(2)-Cl(2B)	96.27(2)
N(2A)-Ti(1)-N(1A)	78.09(3)	N(2B)-Ti(2)-Cl(2B)	89.09(2)
Cl(2A)-Ti(1)-N(1A)	168.67(2)	N(1B)-Ti(2)-Cl(2B)	169.72(2)
O(2A)-Ti(1)-Cl(1A)	91.31(2)	O(1B)-Ti(2)-Cl(1B)	91.83(2)
O(1A)-Ti(1)-Cl(1A)	91.50(2)	O(2B)-Ti(2)-Cl(1B)	90.55(2)
N(2A)-Ti(1)-Cl(1A)	167.43(2)	N(2B)-Ti(2)-Cl(1B)	169.63(2)
Cl(2A)-Ti(1)-Cl(1A)	101.947(12)	N(1B)-Ti(2)-Cl(1B)	89.02(2)
N(1A)-Ti(1)-Cl(1A)	89.36(2)	Cl(2B)-Ti(2)-Cl(1B)	101.259(12)

Table 4. Bond lengths [Å] and angles [°] for IAT98 (CCDC 856446).

Ti(1)-O(2A)	1.8119(7)	Ti(2)-O(2B)	1.8391(7)
Ti(1)-O(1A)	1.8426(7)	Ti(2)-N(2B)	2.2400(9)
Ti(1)-N(2A)	2.2533(9)	Ti(2)-N(1B)	2.2536(9)
Ti(1)-Cl(2A)	2.2832(3)	Ti(2)-Cl(2B)	2.2798(3)
Ti(1)-N(1A)	2.2950(8)	Ti(2)-Cl(1B)	2.3149(3)
Ti(1)-Cl(1A)	2.2961(3)	O(1B)-C(1B)	1.3453(12)
O(1A)-C(1A)	1.3445(12)	O(2B)-C(17B)	1.3520(12)
O(2A)-C(17A)	1.3406(11)	N(1B)-C(7B)	1.3670(13)
N(1A)-C(7A)	1.3661(13)	N(1B)-C(11B)	1.3685(12)
N(1A)-C(11A)	1.3698(13)	N(2B)-C(34B)	1.3356(14)
N(2A)-C(38A)	1.3445(14)	N(2B)-C(38B)	1.3434(14)
N(2A)-C(34A)	1.3474(13)	C(1B)-C(6B)	1.3970(14)
C(1A)-C(6A)	1.3981(14)	C(1B)-C(2B)	1.4071(14)
C(1A)-C(2A)	1.4112(14)	C(2B)-C(3B)	1.3913(15)
C(2A)-C(3A)	1.3868(15)	C(2B)-C(18B)	1.5355(15)
C(2A)-C(18A)	1.5412(15)	C(3B)-C(4B)	1.3956(15)
C(3A)-C(4A)	1.4032(14)	C(4B)-C(5B)	1.3888(15)
C(4A)-C(5A)	1.3818(14)	C(4B)-C(22B)	1.5301(16)
C(4A)-C(22A)	1.5289(15)	C(5B)-C(6B)	1.3904(15)
C(5A)-C(6A)	1.4003(15)	C(6B)-C(7B)	1.4859(14)
C(6A)-C(7A)	1.4833(14)	C(7B)-C(8B)	1.3887(14)
C(7A)-C(8A)	1.3939(14)	C(8B)-C(9B)	1.3775(14)
C(8A)-C(9A)	1.3760(15)	C(9B)-C(10B)	1.3715(15)
C(9A)-C(10A)	1.3786(15)	C(10B)-C(11B)	1.3905(14)
C(10A)-C(11A)	1.3939(14)	C(11B)-C(12B)	1.4823(15)
C(11A)-C(12A)	1.4910(14)	C(12B)-C(17B)	1.3853(15)
C(12A)-C(17A)	1.3940(14)	C(12B)-C(13B)	1.4053(14)
C(12A)-C(13A)	1.4028(14)	C(13B)-C(14B)	1.3806(15)
C(13A)-C(14A)	1.3829(14)	C(14B)-C(15B)	1.3952(16)
C(14A)-C(15A)	1.4008(14)	C(14B)-C(26B)	1.5372(15)
C(14A)-C(26A)	1.5336(14)	C(15B)-C(16B)	1.3918(14)
C(15A)-C(16A)	1.3827(13)	C(16B)-C(17B)	1.4122(15)
C(16A)-C(17A)	1.4099(14)	C(16B)-C(30B)	1.5298(15)
C(16A)-C(30A)	1.5382(14)	C(18B)-C(19B)	1.5328(16)
C(18A)-C(20A)	1.5306(17)	C(18B)-C(21B)	1.5377(15)
C(18A)-C(21A)	1.5340(16)	C(18B)-C(20B)	1.5382(16)
C(18A)-C(19A)	1.5365(16)	C(22B)-C(23B)	1.5194(18)
C(22A)-C(25A)	1.5193(16)	C(22B)-C(24B)	1.5214(17)
C(22A)-C(23A)	1.5292(16)	C(22B)-C(25B)	1.547(2)
C(22A)-C(24A)	1.5466(17)	C(26B)-C(27B)	1.5284(17)
C(26A)-C(29A)	1.5297(17)	C(26B)-C(28B)	1.5301(16)
C(26A)-C(27A)	1.5324(15)	C(26B)-C(29B)	1.5434(16)
C(26A)-C(28A)	1.5352(16)	C(30B)-C(33B)	1.5333(15)
C(30A)-C(32A)	1.5286(14)	C(30B)-C(32B)	1.5395(15)
C(30A)-C(33A)	1.5326(15)	C(30B)-C(31B)	1.5395(15)
C(30A)-C(31A)	1.5381(15)	C(34B)-C(35B)	1.3806(16)
C(34A)-C(35A)	1.3808(15)	C(35B)-C(36B)	1.3711(18)
C(35A)-C(36A)	1.3741(17)	C(36B)-C(37B)	1.3676(18)
C(36A)-C(37A)	1.3860(16)	C(37B)-C(38B)	1.3762(16)
C(37A)-C(38A)	1.3744(15)	C(41)-C(42)	1.358(3)
Ti(2)-O(1B)	1.8294(7)	C(41)-C(46)	1.398(3)

C(42)-C(43)	1.361(2)	C(6A)-C(1A)-C(2A)	121.16(10)
C(43)-C(44)	1.361(2)	C(3A)-C(2A)-C(1A)	116.71(9)
C(44)-C(45)	1.359(2)	C(3A)-C(2A)-C(18A)	121.92(9)
C(45)-C(46)	1.396(3)	C(1A)-C(2A)-C(18A)	121.36(10)
C(51)-C(56)	1.387(4)	C(2A)-C(3A)-C(4A)	124.16(10)
C(51)-C(52)	1.388(3)	C(5A)-C(4A)-C(3A)	116.92(10)
C(52)-C(53)	1.350(4)	C(5A)-C(4A)-C(22A)	123.17(9)
C(53)-C(54)	1.388(4)	C(3A)-C(4A)-C(22A)	119.91(9)
C(54)-C(55)	1.388(3)	C(4A)-C(5A)-C(6A)	121.94(9)
C(55)-C(56)	1.367(4)	C(1A)-C(6A)-C(5A)	119.05(9)
C(61)-C(62)	1.3647(19)	C(1A)-C(6A)-C(7A)	122.17(9)
C(61)-C(66)	1.3661(19)	C(5A)-C(6A)-C(7A)	118.71(9)
C(62)-C(63)	1.3783(19)	N(1A)-C(7A)-C(8A)	121.27(9)
C(63)-C(64)	1.3705(19)	N(1A)-C(7A)-C(6A)	121.22(9)
C(64)-C(65)	1.374(2)	C(8A)-C(7A)-C(6A)	117.48(9)
C(65)-C(66)	1.375(2)	C(9A)-C(8A)-C(7A)	120.19(10)
C(71)-C(72)	1.3648(18)	C(8A)-C(9A)-C(10A)	118.44(10)
C(71)-C(76)	1.3843(18)	C(9A)-C(10A)-C(11A)	120.20(10)
C(72)-C(73)	1.3777(19)	N(1A)-C(11A)-C(10A)	121.06(9)
C(73)-C(74)	1.3844(19)	N(1A)-C(11A)-C(12A)	121.07(9)
C(74)-C(75)	1.3714(18)	C(10A)-C(11A)-C(12A)	117.83(9)
C(75)-C(76)	1.3794(18)	C(17A)-C(12A)-C(13A)	118.33(9)
C(81)-C(86)	1.374(2)	C(17A)-C(12A)-C(11A)	122.85(9)
C(81)-C(82)	1.426(2)	C(13A)-C(12A)-C(11A)	118.83(9)
C(82)-C(83)	1.371(2)	C(14A)-C(13A)-C(12A)	121.49(9)
C(83)-C(84)	1.345(2)	C(13A)-C(14A)-C(15A)	117.80(9)
C(84)-C(85)	1.3426(18)	C(13A)-C(14A)-C(26A)	123.07(9)
C(85)-C(86)	1.3461(18)	C(15A)-C(14A)-C(26A)	119.13(9)
		C(16A)-C(15A)-C(14A)	123.71(9)
O(2A)-Ti(1)-O(1A)	166.32(3)	C(15A)-C(16A)-C(17A)	116.33(9)
O(2A)-Ti(1)-N(2A)	86.38(3)	C(15A)-C(16A)-C(30A)	122.58(9)
O(1A)-Ti(1)-N(2A)	87.98(3)	C(17A)-C(16A)-C(30A)	121.08(9)
O(2A)-Ti(1)-Cl(2A)	96.07(2)	O(2A)-C(17A)-C(12A)	118.18(9)
O(1A)-Ti(1)-Cl(2A)	96.43(2)	O(2A)-C(17A)-C(16A)	119.46(9)
N(2A)-Ti(1)-Cl(2A)	90.59(2)	C(12A)-C(17A)-C(16A)	122.33(9)
O(2A)-Ti(1)-N(1A)	82.89(3)	C(20A)-C(18A)-C(21A)	107.71(9)
O(1A)-Ti(1)-N(1A)	83.75(3)	C(20A)-C(18A)-C(19A)	109.76(10)
N(2A)-Ti(1)-N(1A)	78.09(3)	C(21A)-C(18A)-C(19A)	108.04(10)
Cl(2A)-Ti(1)-N(1A)	168.67(2)	C(20A)-C(18A)-C(2A)	109.97(9)
O(2A)-Ti(1)-Cl(1A)	91.31(2)	C(21A)-C(18A)-C(2A)	111.22(9)
O(1A)-Ti(1)-Cl(1A)	91.50(2)	C(19A)-C(18A)-C(2A)	110.09(9)
N(2A)-Ti(1)-Cl(1A)	167.43(2)	C(25A)-C(22A)-C(23A)	108.97(10)
Cl(2A)-Ti(1)-Cl(1A)	101.947(12)	C(25A)-C(22A)-C(4A)	112.20(9)
N(1A)-Ti(1)-Cl(1A)	89.36(2)	C(23A)-C(22A)-C(4A)	109.90(9)
C(1A)-O(1A)-Ti(1)	137.07(6)	C(25A)-C(22A)-C(24A)	108.01(10)
C(17A)-O(2A)-Ti(1)	143.37(6)	C(23A)-C(22A)-C(24A)	108.37(10)
C(7A)-N(1A)-C(11A)	117.77(8)	C(4A)-C(22A)-C(24A)	109.31(9)
C(7A)-N(1A)-Ti(1)	118.19(6)	C(29A)-C(26A)-C(27A)	108.85(9)
C(11A)-N(1A)-Ti(1)	119.23(7)	C(29A)-C(26A)-C(14A)	109.19(10)
C(38A)-N(2A)-C(34A)	117.16(9)	C(27A)-C(26A)-C(14A)	112.19(9)
C(38A)-N(2A)-Ti(1)	121.00(7)	C(29A)-C(26A)-C(28A)	109.25(10)
C(34A)-N(2A)-Ti(1)	121.73(7)	C(27A)-C(26A)-C(28A)	108.02(10)
O(1A)-C(1A)-C(6A)	117.68(9)	C(14A)-C(26A)-C(28A)	109.29(9)
O(1A)-C(1A)-C(2A)	121.13(9)	C(32A)-C(30A)-C(33A)	107.49(8)

C(32A)-C(30A)-C(31A)	107.54(9)	N(1B)-C(11B)-C(12B)	120.20(9)
C(33A)-C(30A)-C(31A)	109.80(9)	C(10B)-C(11B)-C(12B)	118.70(9)
C(32A)-C(30A)-C(16A)	111.17(8)	C(17B)-C(12B)-C(13B)	119.38(10)
C(33A)-C(30A)-C(16A)	110.01(9)	C(17B)-C(12B)-C(11B)	122.47(9)
C(31A)-C(30A)-C(16A)	110.75(8)	C(13B)-C(12B)-C(11B)	118.14(10)
N(2A)-C(34A)-C(35A)	122.65(11)	C(14B)-C(13B)-C(12B)	121.00(10)
C(36A)-C(35A)-C(34A)	119.43(11)	C(13B)-C(14B)-C(15B)	117.79(10)
C(35A)-C(36A)-C(37A)	118.55(11)	C(13B)-C(14B)-C(26B)	123.17(10)
C(38A)-C(37A)-C(36A)	118.88(11)	C(15B)-C(14B)-C(26B)	119.03(10)
N(2A)-C(38A)-C(37A)	123.31(10)	C(16B)-C(15B)-C(14B)	123.93(10)
O(1B)-Ti(2)-O(2B)	167.35(3)	C(15B)-C(16B)-C(17B)	116.20(10)
O(1B)-Ti(2)-N(2B)	87.79(3)	C(15B)-C(16B)-C(30B)	121.76(10)
O(2B)-Ti(2)-N(2B)	87.65(3)	C(17B)-C(16B)-C(30B)	122.01(9)
O(1B)-Ti(2)-N(1B)	83.83(3)	O(2B)-C(17B)-C(12B)	117.57(9)
O(2B)-Ti(2)-N(1B)	83.80(3)	O(2B)-C(17B)-C(16B)	120.75(9)
N(2B)-Ti(2)-N(1B)	80.64(3)	C(12B)-C(17B)-C(16B)	121.66(9)
O(1B)-Ti(2)-Cl(2B)	95.44(2)	C(19B)-C(18B)-C(2B)	111.91(10)
O(2B)-Ti(2)-Cl(2B)	96.27(2)	C(19B)-C(18B)-C(21B)	107.34(9)
N(2B)-Ti(2)-Cl(2B)	89.09(2)	C(2B)-C(18B)-C(21B)	110.13(8)
N(1B)-Ti(2)-Cl(2B)	169.72(2)	C(19B)-C(18B)-C(20B)	107.46(9)
O(1B)-Ti(2)-Cl(1B)	91.83(2)	C(2B)-C(18B)-C(20B)	110.03(9)
O(2B)-Ti(2)-Cl(1B)	90.55(2)	C(21B)-C(18B)-C(20B)	109.90(10)
N(2B)-Ti(2)-Cl(1B)	169.63(2)	C(23B)-C(22B)-C(24B)	108.82(11)
N(1B)-Ti(2)-Cl(1B)	89.02(2)	C(23B)-C(22B)-C(4B)	112.22(11)
Cl(2B)-Ti(2)-Cl(1B)	101.259(12)	C(24B)-C(22B)-C(4B)	110.33(10)
C(1B)-O(1B)-Ti(2)	139.65(6)	C(23B)-C(22B)-C(25B)	108.42(12)
C(17B)-O(2B)-Ti(2)	136.50(7)	C(24B)-C(22B)-C(25B)	108.41(11)
C(7B)-N(1B)-C(11B)	117.44(8)	C(4B)-C(22B)-C(25B)	108.55(11)
C(7B)-N(1B)-Ti(2)	119.11(6)	C(27B)-C(26B)-C(28B)	108.86(10)
C(11B)-N(1B)-Ti(2)	119.40(7)	C(27B)-C(26B)-C(14B)	112.26(9)
C(34B)-N(2B)-C(38B)	117.01(9)	C(28B)-C(26B)-C(14B)	109.38(9)
C(34B)-N(2B)-Ti(2)	122.29(7)	C(27B)-C(26B)-C(29B)	108.14(9)
C(38B)-N(2B)-Ti(2)	120.70(7)	C(28B)-C(26B)-C(29B)	109.36(10)
O(1B)-C(1B)-C(6B)	117.12(9)	C(14B)-C(26B)-C(29B)	108.79(9)
O(1B)-C(1B)-C(2B)	121.21(9)	C(16B)-C(30B)-C(33B)	111.82(9)
C(6B)-C(1B)-C(2B)	121.66(10)	C(16B)-C(30B)-C(32B)	111.64(9)
C(3B)-C(2B)-C(1B)	116.27(10)	C(33B)-C(30B)-C(32B)	107.29(9)
C(3B)-C(2B)-C(18B)	122.50(9)	C(16B)-C(30B)-C(31B)	108.49(9)
C(1B)-C(2B)-C(18B)	121.23(10)	C(33B)-C(30B)-C(31B)	107.87(9)
C(2B)-C(3B)-C(4B)	124.03(10)	C(32B)-C(30B)-C(31B)	109.66(9)
C(5B)-C(4B)-C(3B)	117.27(10)	N(2B)-C(34B)-C(35B)	123.03(11)
C(5B)-C(4B)-C(22B)	119.09(10)	C(36B)-C(35B)-C(34B)	119.10(12)
C(3B)-C(4B)-C(22B)	123.60(10)	C(37B)-C(36B)-C(35B)	118.62(11)
C(4B)-C(5B)-C(6B)	121.60(10)	C(36B)-C(37B)-C(38B)	119.34(12)
C(5B)-C(6B)-C(1B)	119.06(9)	N(2B)-C(38B)-C(37B)	122.90(11)
C(5B)-C(6B)-C(7B)	118.33(9)	C(42)-C(41)-C(46)	119.66(17)
C(1B)-C(6B)-C(7B)	122.60(9)	C(41)-C(42)-C(43)	120.35(17)
N(1B)-C(7B)-C(8B)	121.50(9)	C(44)-C(43)-C(42)	121.23(17)
N(1B)-C(7B)-C(6B)	120.24(9)	C(45)-C(44)-C(43)	119.78(16)
C(8B)-C(7B)-C(6B)	118.06(9)	C(44)-C(45)-C(46)	120.22(18)
C(9B)-C(8B)-C(7B)	119.77(10)	C(45)-C(46)-C(41)	118.77(18)
C(10B)-C(9B)-C(8B)	118.64(10)	C(56)-C(51)-C(52)	118.9(3)
C(9B)-C(10B)-C(11B)	120.32(10)	C(53)-C(52)-C(51)	119.7(3)
N(1B)-C(11B)-C(10B)	121.00(10)	C(52)-C(53)-C(54)	121.4(2)

C(55)-C(54)-C(53)	119.6(3)
C(56)-C(55)-C(54)	118.6(3)
C(55)-C(56)-C(51)	121.8(2)
C(62)-C(61)-C(66)	120.00(12)
C(61)-C(62)-C(63)	120.17(12)
C(64)-C(63)-C(62)	120.08(13)
C(63)-C(64)-C(65)	119.44(12)
C(64)-C(65)-C(66)	120.25(13)
C(61)-C(66)-C(65)	120.06(13)
C(72)-C(71)-C(76)	119.99(13)
C(71)-C(72)-C(73)	120.19(12)
C(72)-C(73)-C(74)	119.88(13)
C(75)-C(74)-C(73)	120.08(13)
C(74)-C(75)-C(76)	119.72(12)
C(75)-C(76)-C(71)	120.10(13)
C(86)-C(81)-C(82)	118.51(13)
C(83)-C(82)-C(81)	118.44(13)
C(84)-C(83)-C(82)	120.46(13)
C(85)-C(84)-C(83)	121.16(14)
C(84)-C(85)-C(86)	121.11(14)
C(85)-C(86)-C(81)	120.32(13)

Table 5. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^4$) for IAT98 (CCDC 856446). The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
Ti(1)	102(1)	100(1)	127(1)	4(1)	20(1)	-2(1)
Cl(1A)	209(1)	210(1)	139(1)	-34(1)	14(1)	8(1)
Cl(2A)	149(1)	154(1)	254(1)	9(1)	47(1)	-40(1)
O(1A)	113(4)	119(4)	168(4)	22(3)	13(3)	-2(3)
O(2A)	116(4)	97(3)	146(4)	4(3)	22(3)	-2(3)
N(1A)	104(4)	101(4)	133(4)	7(3)	17(3)	11(3)
N(2A)	142(5)	111(4)	159(4)	-11(3)	33(4)	-15(4)
C(1A)	110(5)	108(5)	151(5)	-22(4)	21(4)	19(4)
C(2A)	147(6)	104(5)	172(5)	7(4)	35(4)	16(4)
C(3A)	167(6)	117(5)	169(5)	28(4)	22(4)	34(4)
C(4A)	124(5)	133(5)	172(5)	6(4)	21(4)	15(4)
C(5A)	121(5)	122(5)	182(5)	11(4)	43(4)	-1(4)
C(6A)	122(5)	99(5)	150(5)	4(4)	29(4)	26(4)
C(7A)	114(5)	107(5)	155(5)	-2(4)	23(4)	-1(4)
C(8A)	122(5)	190(6)	220(6)	46(5)	58(5)	44(4)
C(9A)	154(6)	228(6)	216(6)	65(5)	94(5)	40(5)
C(10A)	167(6)	165(5)	193(6)	70(4)	57(5)	34(4)
C(11A)	124(5)	110(5)	145(5)	10(4)	19(4)	7(4)
C(12A)	114(5)	125(5)	113(5)	7(4)	18(4)	22(4)
C(13A)	132(5)	136(5)	160(5)	21(4)	44(4)	2(4)
C(14A)	148(5)	123(5)	139(5)	12(4)	27(4)	28(4)
C(15A)	127(5)	144(5)	125(5)	2(4)	16(4)	45(4)
C(16A)	109(5)	131(5)	90(5)	-18(4)	3(4)	12(4)
C(17A)	133(5)	101(5)	92(5)	-7(4)	11(4)	7(4)
C(18A)	170(6)	147(5)	267(6)	85(5)	29(5)	-14(5)
C(19A)	236(7)	299(7)	285(7)	94(5)	100(6)	-37(5)
C(20A)	264(7)	150(6)	419(8)	28(5)	34(6)	-57(5)
C(21A)	254(7)	303(7)	480(9)	246(6)	8(6)	-48(6)
C(22A)	130(6)	199(6)	236(6)	44(5)	-7(5)	3(5)
C(23A)	221(7)	388(8)	424(8)	41(6)	-38(6)	82(6)
C(24A)	258(7)	402(8)	289(7)	-9(6)	-51(6)	-25(6)
C(25A)	196(7)	485(9)	456(9)	202(7)	-62(6)	-70(6)
C(26A)	176(6)	143(5)	279(6)	90(5)	83(5)	52(4)
C(27A)	193(6)	151(6)	369(7)	95(5)	74(5)	16(5)
C(28A)	500(9)	109(6)	492(9)	32(6)	257(7)	57(6)
C(29A)	194(7)	383(8)	404(8)	253(6)	84(6)	68(6)
C(30A)	94(5)	150(5)	158(5)	3(4)	20(4)	4(4)
C(31A)	127(6)	186(6)	261(6)	-19(5)	5(5)	-22(4)
C(32A)	126(5)	201(6)	239(6)	12(5)	43(5)	29(5)
C(33A)	172(6)	236(6)	189(6)	29(5)	70(5)	25(5)
C(34A)	157(6)	149(5)	216(6)	-17(4)	46(5)	-12(4)
C(35A)	216(6)	186(6)	294(7)	-70(5)	137(5)	-28(5)
C(36A)	317(7)	197(6)	208(6)	-66(5)	132(5)	-93(5)
C(37A)	318(7)	181(6)	164(6)	-9(4)	29(5)	-40(5)
C(38A)	222(6)	146(5)	176(6)	-11(4)	27(5)	-3(5)

Ti(2)	106(1)	105(1)	127(1)	-10(1)	9(1)	-12(1)
Cl(1B)	138(1)	199(1)	189(1)	-35(1)	47(1)	-16(1)
Cl(2B)	165(1)	173(1)	155(1)	-31(1)	6(1)	-52(1)
O(1B)	121(4)	114(3)	150(4)	-11(3)	0(3)	-9(3)
O(2B)	134(4)	105(3)	171(4)	9(3)	1(3)	-28(3)
N(1B)	92(4)	109(4)	153(4)	-15(3)	12(3)	-3(3)
N(2B)	128(5)	152(4)	163(5)	-28(4)	10(4)	-21(4)
C(1B)	107(5)	145(5)	124(5)	7(4)	31(4)	-9(4)
C(2B)	141(5)	125(5)	161(5)	-4(4)	51(4)	8(4)
C(3B)	192(6)	158(5)	184(6)	12(4)	34(5)	58(5)
C(4B)	198(6)	206(6)	151(5)	-8(4)	-1(5)	63(5)
C(5B)	172(6)	179(6)	147(5)	-38(4)	18(4)	12(4)
C(6B)	125(5)	136(5)	126(5)	-3(4)	34(4)	14(4)
C(7B)	92(5)	138(5)	154(5)	-26(4)	6(4)	-8(4)
C(8B)	156(6)	179(6)	151(5)	-33(4)	2(4)	6(4)
C(9B)	191(6)	191(6)	191(6)	-77(5)	18(5)	31(5)
C(10B)	172(6)	126(5)	226(6)	-40(4)	-1(5)	26(4)
C(11B)	90(5)	125(5)	194(5)	-25(4)	-4(4)	-4(4)
C(12B)	109(5)	110(5)	200(6)	-7(4)	15(4)	10(4)
C(13B)	140(6)	124(5)	243(6)	-39(4)	-4(5)	3(4)
C(14B)	130(5)	120(5)	275(6)	7(5)	31(5)	3(4)
C(15B)	148(6)	152(5)	231(6)	48(4)	28(5)	5(4)
C(16B)	112(5)	131(5)	205(6)	25(4)	20(4)	7(4)
C(17B)	102(5)	106(5)	210(6)	3(4)	23(4)	0(4)
C(18B)	180(6)	115(5)	228(6)	-27(4)	30(5)	-9(4)
C(19B)	327(8)	142(6)	383(8)	-34(5)	-13(6)	28(5)
C(20B)	216(6)	182(6)	335(7)	-45(5)	70(5)	-58(5)
C(21B)	272(7)	183(6)	244(6)	-84(5)	71(5)	-31(5)
C(22B)	329(8)	296(7)	233(7)	-72(5)	-112(6)	142(6)
C(23B)	779(13)	421(9)	536(10)	-160(8)	-409(9)	347(9)
C(24B)	450(9)	334(7)	192(6)	-5(5)	-98(6)	75(6)
C(25B)	215(8)	760(11)	431(9)	-217(8)	-78(7)	56(8)
C(26B)	183(6)	102(5)	328(7)	15(5)	30(5)	-22(4)
C(27B)	277(7)	156(6)	415(8)	-7(5)	28(6)	-62(5)
C(28B)	238(7)	169(6)	452(8)	15(5)	99(6)	-51(5)
C(29B)	252(7)	137(6)	451(8)	47(5)	55(6)	2(5)
C(30B)	149(6)	178(6)	185(6)	37(4)	-5(4)	-26(4)
C(31B)	143(6)	214(6)	243(6)	32(5)	-21(5)	-3(5)
C(32B)	172(6)	225(6)	185(6)	-5(5)	21(5)	-36(5)
C(33B)	289(7)	254(7)	221(6)	69(5)	-12(5)	-61(5)
C(34B)	215(6)	236(6)	267(6)	38(5)	82(5)	25(5)
C(35B)	305(8)	410(8)	312(7)	40(6)	177(6)	2(6)
C(36B)	180(7)	454(8)	345(8)	-106(6)	121(6)	-16(6)
C(37B)	175(7)	444(8)	329(7)	-40(6)	33(6)	108(6)
C(38B)	176(6)	333(7)	224(6)	10(5)	14(5)	70(5)
C(41)	768(15)	614(12)	625(13)	-185(10)	356(12)	-219(11)
C(42)	716(13)	438(10)	418(10)	-107(8)	155(9)	34(9)
C(43)	505(10)	516(10)	371(9)	-149(8)	105(8)	7(8)
C(44)	567(12)	518(10)	417(9)	-49(8)	154(9)	-7(9)
C(45)	687(14)	705(13)	507(11)	53(9)	95(10)	264(11)
C(46)	373(11)	1098(17)	792(15)	-337(14)	171(11)	24(12)

C(51)	1110(20)	1210(20)	1250(20)	732(18)	854(18)	582(17)
C(52)	980(20)	1950(30)	462(13)	287(15)	292(14)	810(20)
C(53)	691(16)	2530(30)	315(11)	26(16)	150(11)	920(20)
C(54)	830(16)	1990(30)	444(12)	359(15)	310(12)	828(17)
C(55)	1270(20)	1187(19)	1015(18)	650(15)	905(18)	750(17)
C(56)	1420(30)	934(18)	1790(30)	770(19)	1400(20)	700(18)
C(61)	243(7)	224(7)	547(9)	-59(6)	154(7)	-40(6)
C(62)	428(9)	255(7)	301(7)	-20(6)	159(7)	-116(6)
C(63)	289(8)	238(7)	471(9)	85(6)	-45(7)	-39(6)
C(64)	259(7)	240(7)	663(11)	-33(7)	241(7)	-30(6)
C(65)	552(10)	348(8)	310(8)	1(6)	211(7)	-89(7)
C(66)	353(8)	303(7)	419(8)	47(6)	-40(7)	-2(6)
C(71)	195(7)	443(8)	358(8)	-146(6)	110(6)	-71(6)
C(72)	178(7)	618(10)	234(7)	-112(7)	20(5)	59(6)
C(73)	332(8)	544(10)	273(7)	92(7)	48(6)	158(7)
C(74)	249(7)	335(8)	401(8)	0(6)	7(6)	82(6)
C(75)	154(6)	413(8)	275(7)	-103(6)	4(5)	24(6)
C(76)	254(7)	476(9)	267(7)	-53(6)	119(6)	-52(6)
C(81)	675(12)	243(8)	429(9)	55(6)	-201(8)	21(8)
C(82)	226(8)	585(11)	689(12)	-289(9)	-4(8)	80(8)
C(83)	452(10)	418(9)	434(9)	-77(7)	199(8)	-214(8)
C(84)	484(10)	256(7)	316(8)	-10(6)	43(7)	-62(7)
C(85)	323(8)	368(8)	306(7)	-57(6)	33(6)	-13(6)
C(86)	399(9)	419(8)	232(7)	33(6)	48(6)	-130(7)

CALIFORNIA INSTITUTE OF TECHNOLOGY
BECKMAN INSTITUTE
X-RAY CRYSTALLOGRAPHY LABORATORY



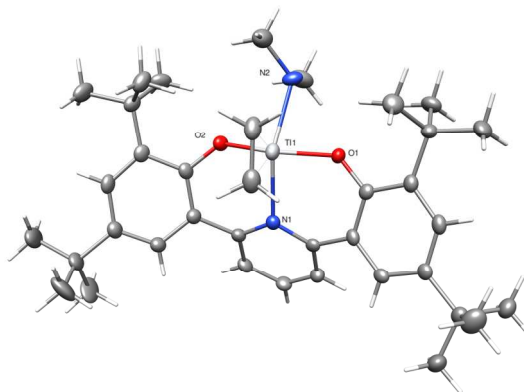
Date 27 April 2011

Crystal Structure Analysis of:
Complex 7 (ONO)Ti(HNMe₂)(C₂H₄) (IAT87)
(shown below)

By Michael W. Day 116 Beckman ext. 2734
e-mail: mikeday@caltech.edu

Contents

Table 1. Crystal data
Figures Minimum overlap
Table 2. Atomic Coordinates
Table 3. Selected bond distances and angles
Table 4. Full bond distances and angles
Table 5. Anisotropic displacement parameters
Table 6. Observed and calculated structure factors (available upon request)



IAT87

Note: The crystallographic data have been deposited in the Cambridge Database (CCDC). The deposition number is 822994.

Table 1. Crystal data and structure refinement for IAT87 (CCDC 822994).

Empirical formula	$C_{37}H_{53}N_2O_2Ti \cdot 0.5(C_5H_{12})$
Formula weight	641.79
Crystallization Solvent	Pentane
Crystal Habit	Blade
Crystal size	0.21 x 0.13 x 0.08 mm ³
Crystal color	Dark red



Data Collection

Type of diffractometer	Bruker KAPPA APEX II	
Wavelength	0.71073 Å MoK α	
Data Collection Temperature	100(2) K	
θ range for 9906 reflections used in lattice determination	2.18 to 25.01°	
Unit cell dimensions	a = 17.7536(5) Å b = 10.5239(3) Å c = 20.7880(6) Å	$\alpha = 90^\circ$ $\beta = 104.343(2)^\circ$ $\gamma = 90^\circ$
Volume	3762.91(19) Å ³	
Z	4	
Crystal system	Monoclinic	
Space group	P 2 ₁ /n	
Density (calculated)	1.133 Mg/m ³	
F(000)	1392	
θ range for data collection	2.02 to 29.99°	
Completeness to $\theta = 29.99^\circ$	93.2 %	
Index ranges	-23 \leq h \leq 24, -14 \leq k \leq 14, -28 \leq l \leq 29	
Data collection scan type	ω scans; 14 settings	
Reflections collected	109497	
Independent reflections	10215 [$R_{int} = 0.0878$]	
Absorption coefficient	0.261 mm ⁻¹	
Absorption correction	None	
Max. and min. transmission	0.9794 and 0.9472	

Table 1 (cont.)**Structure solution and Refinement**

Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on F^2
Data / restraints / parameters	10215 / 7 / 464
Treatment of hydrogen atoms	Riding
Goodness-of-fit on F^2	2.095
Final R indices [$I > 2\sigma(I)$, 5835 reflections]	$R1 = 0.0612$, $wR2 = 0.0614$
R indices (all data)	$R1 = 0.1167$, $wR2 = 0.0627$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.003
Average shift/error	0.000
Largest diff. peak and hole	1.294 and -0.596 e.Å ⁻³

Special Refinement Details

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

One t-Bu is disordered by the common rotation around the central carbon and was included in refinement as both positions with a total occupancy of one. Pentane is included as a solvent of crystallization and sits near a center of symmetry and bridges that center. It was included as a restrained pentane at half occupancy.

Refinement of F^2 against ALL reflections. The weighted R-factor (wR) and goodness of fit (S) are based on F^2 , conventional R-factors (R) are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

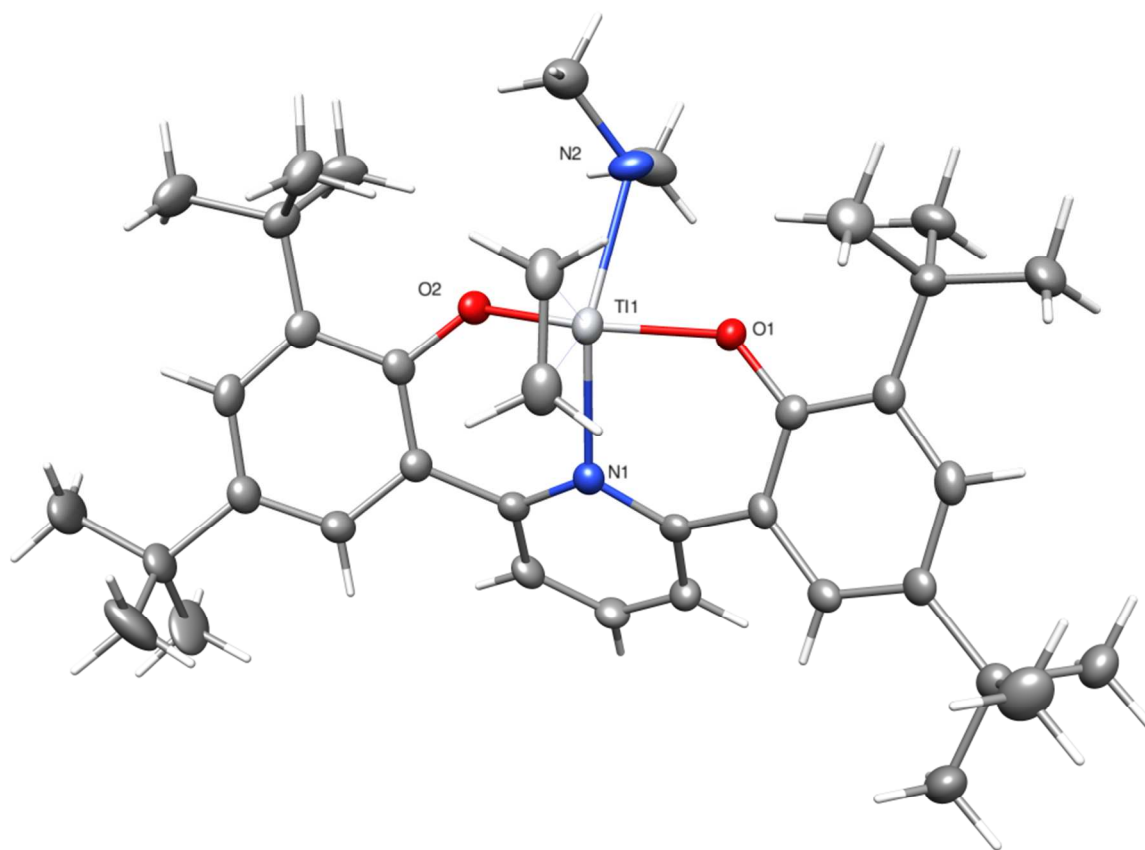


Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for IAT87 (CCDC 822994). U_{eq} is defined as the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U_{eq}	Occ
Ti(1)	6825(1)	1750(1)	-437(1)	26(1)	1
O(1)	7089(1)	1327(1)	-1243(1)	24(1)	1
O(2)	6142(1)	1328(1)	100(1)	26(1)	1
N(1)	7463(1)	-17(2)	-36(1)	22(1)	1
N(2)	5779(1)	2486(2)	-1209(1)	31(1)	1
C(1)	7798(1)	986(2)	-1315(1)	23(1)	1
C(2)	8099(1)	1520(2)	-1820(1)	25(1)	1
C(3)	8832(1)	1134(2)	-1852(1)	26(1)	1
C(4)	9288(1)	244(2)	-1425(1)	24(1)	1
C(5)	8966(1)	-283(2)	-950(1)	24(1)	1
C(6)	8219(1)	58(2)	-890(1)	22(1)	1
C(7)	7892(1)	-637(2)	-403(1)	22(1)	1
C(8)	8037(1)	-1944(2)	-320(1)	27(1)	1
C(9)	7782(1)	-2619(2)	146(1)	27(1)	1
C(10)	7404(1)	-1973(2)	547(1)	27(1)	1
C(11)	7258(1)	-676(2)	463(1)	24(1)	1
C(12)	6881(1)	-27(2)	931(1)	24(1)	1
C(13)	7087(1)	-415(2)	1597(1)	29(1)	1
C(14)	6726(1)	59(2)	2064(1)	31(1)	1
C(15)	6130(1)	940(2)	1839(1)	29(1)	1
C(16)	5897(1)	1375(2)	1189(1)	26(1)	1
C(17)	6311(1)	907(2)	731(1)	24(1)	1
C(18)	7630(1)	2503(2)	-2314(1)	28(1)	1
C(19)	6862(1)	1940(2)	-2699(1)	38(1)	1
C(20)	7471(1)	3687(2)	-1939(1)	36(1)	1
C(21)	8077(1)	2950(2)	-2816(1)	45(1)	1
C(22)	10098(1)	-94(2)	-1503(1)	28(1)	1
C(23)	10600(1)	1106(2)	-1443(1)	48(1)	1
C(24)	10029(1)	-691(2)	-2188(1)	38(1)	1
C(25)	10512(1)	-1037(2)	-981(1)	47(1)	1
C(26)	6981(2)	-340(3)	2800(1)	39(1)	1
C(27A)	7772(3)	382(5)	3108(2)	63(2)	0.701(8)
C(28A)	7116(4)	-1713(7)	2865(3)	60(2)	0.701(8)
C(29A)	6441(3)	139(6)	3206(2)	64(2)	0.701(8)
C(27B)	7619(9)	-1360(20)	2940(7)	97(9)	0.299(8)
C(28B)	6257(6)	-1090(17)	2975(5)	84(7)	0.299(8)
C(29B)	7171(9)	674(11)	3258(5)	71(5)	0.299(8)
C(30)	5222(1)	2315(2)	968(1)	28(1)	1
C(31)	4594(1)	1748(2)	400(1)	39(1)	1
C(32)	4834(1)	2631(2)	1530(1)	44(1)	1
C(33)	5522(1)	3574(2)	747(1)	39(1)	1
C(34)	7756(1)	2804(2)	128(1)	35(1)	1
C(35)	7164(1)	3668(2)	-203(1)	34(1)	1
C(36)	5306(1)	3518(2)	-1054(1)	50(1)	1
C(37)	5267(1)	1399(2)	-1468(1)	49(1)	1

C(41)	9722(8)	5208(8)	8889(4)	371(16)	0.50
C(42)	10005(6)	4068(7)	9349(4)	213(7)	0.50
C(43)	9933(6)	4262(8)	10064(3)	178(6)	0.50
C(44)	10476(5)	5294(9)	10450(4)	136(5)	0.50
C(45)	10312(5)	5626(8)	11124(3)	128(5)	0.50

Table 3. Selected bond lengths [Å] and angles [°] for IAT87 (CCDC 822994).

Ti(1)-O(2)	1.8924(14)	O(2)-Ti(1)-O(1)	143.01(6)
Ti(1)-O(1)	1.9014(14)	O(2)-Ti(1)-C(34)	108.97(8)
Ti(1)-C(34)	2.093(2)	O(1)-Ti(1)-C(34)	106.60(8)
Ti(1)-C(35)	2.128(2)	O(2)-Ti(1)-C(35)	105.92(8)
Ti(1)-N(1)	2.2288(17)	O(1)-Ti(1)-C(35)	107.80(8)
Ti(1)-N(2)	2.2677(17)	C(34)-Ti(1)-C(35)	39.59(7)
		O(2)-Ti(1)-N(1)	85.62(6)
		O(1)-Ti(1)-N(1)	85.47(6)
		C(34)-Ti(1)-N(1)	88.55(8)
		C(35)-Ti(1)-N(1)	128.08(7)
		O(2)-Ti(1)-N(2)	88.05(6)
		O(1)-Ti(1)-N(2)	78.12(6)
		C(34)-Ti(1)-N(2)	127.60(8)
		C(35)-Ti(1)-N(2)	88.49(8)
		N(1)-Ti(1)-N(2)	143.18(7)

Table 4. Bond lengths [Å] and angles [°] for IAT87 (CCDC 822994).

Ti(1)-O(2)	1.8924(14)	C(42)-C(43)	1.5373(10)
Ti(1)-O(1)	1.9014(14)	C(43)-C(44)	1.5397(10)
Ti(1)-C(34)	2.093(2)	C(44)-C(45)	1.5409(10)
Ti(1)-C(35)	2.128(2)		
Ti(1)-N(1)	2.2288(17)	O(2)-Ti(1)-O(1)	143.01(6)
Ti(1)-N(2)	2.2677(17)	O(2)-Ti(1)-C(34)	108.97(8)
O(1)-C(1)	1.351(2)	O(1)-Ti(1)-C(34)	106.60(8)
O(2)-C(17)	1.347(2)	O(2)-Ti(1)-C(35)	105.92(8)
N(1)-C(11)	1.369(3)	O(1)-Ti(1)-C(35)	107.80(8)
N(1)-C(7)	1.370(3)	C(34)-Ti(1)-C(35)	39.59(7)
N(2)-C(36)	1.458(2)	O(2)-Ti(1)-N(1)	85.62(6)
N(2)-C(37)	1.478(2)	O(1)-Ti(1)-N(1)	85.47(6)
C(1)-C(6)	1.401(3)	C(34)-Ti(1)-N(1)	88.55(8)
C(1)-C(2)	1.410(3)	C(35)-Ti(1)-N(1)	128.08(7)
C(2)-C(3)	1.380(3)	O(2)-Ti(1)-N(2)	88.05(6)
C(2)-C(18)	1.547(3)	O(1)-Ti(1)-N(2)	78.12(6)
C(3)-C(4)	1.401(3)	C(34)-Ti(1)-N(2)	127.60(8)
C(4)-C(5)	1.375(3)	C(35)-Ti(1)-N(2)	88.49(8)
C(4)-C(22)	1.529(3)	N(1)-Ti(1)-N(2)	143.18(7)
C(5)-C(6)	1.407(3)	C(1)-O(1)-Ti(1)	126.81(13)
C(6)-C(7)	1.481(3)	C(17)-O(2)-Ti(1)	129.15(13)
C(7)-C(8)	1.403(3)	C(11)-N(1)-C(7)	117.98(19)
C(8)-C(9)	1.365(3)	C(11)-N(1)-Ti(1)	120.03(15)
C(9)-C(10)	1.373(3)	C(7)-N(1)-Ti(1)	119.33(15)
C(10)-C(11)	1.393(3)	C(36)-N(2)-C(37)	108.71(18)
C(11)-C(12)	1.478(3)	C(36)-N(2)-Ti(1)	120.91(14)
C(12)-C(17)	1.397(3)	C(37)-N(2)-Ti(1)	108.43(13)
C(12)-C(13)	1.402(3)	O(1)-C(1)-C(6)	118.8(2)
C(13)-C(14)	1.384(3)	O(1)-C(1)-C(2)	120.8(2)
C(14)-C(15)	1.398(3)	C(6)-C(1)-C(2)	120.3(2)
C(14)-C(26)	1.542(3)	C(3)-C(2)-C(1)	117.0(2)
C(15)-C(16)	1.389(3)	C(3)-C(2)-C(18)	121.6(2)
C(16)-C(17)	1.426(3)	C(1)-C(2)-C(18)	121.3(2)
C(16)-C(30)	1.534(3)	C(2)-C(3)-C(4)	124.7(2)
C(18)-C(19)	1.520(3)	C(5)-C(4)-C(3)	116.6(2)
C(18)-C(21)	1.533(3)	C(5)-C(4)-C(22)	123.5(2)
C(18)-C(20)	1.533(3)	C(3)-C(4)-C(22)	119.9(2)
C(22)-C(25)	1.518(3)	C(4)-C(5)-C(6)	121.8(2)
C(22)-C(24)	1.533(3)	C(1)-C(6)-C(5)	119.4(2)
C(22)-C(23)	1.533(3)	C(1)-C(6)-C(7)	122.4(2)
C(26)-C(29B)	1.414(11)	C(5)-C(6)-C(7)	118.2(2)
C(26)-C(28A)	1.466(7)	N(1)-C(7)-C(8)	120.7(2)
C(26)-C(29A)	1.512(4)	N(1)-C(7)-C(6)	120.9(2)
C(26)-C(27B)	1.535(17)	C(8)-C(7)-C(6)	118.5(2)
C(26)-C(27A)	1.584(5)	C(9)-C(8)-C(7)	120.9(2)
C(26)-C(28B)	1.625(10)	C(8)-C(9)-C(10)	118.1(2)
C(30)-C(31)	1.530(3)	C(9)-C(10)-C(11)	120.8(2)
C(30)-C(32)	1.532(3)	N(1)-C(11)-C(10)	121.0(2)
C(30)-C(33)	1.540(3)	N(1)-C(11)-C(12)	120.9(2)
C(34)-C(35)	1.430(2)	C(10)-C(11)-C(12)	118.1(2)
C(41)-C(42)	1.5389(10)	C(17)-C(12)-C(13)	119.4(2)

C(17)-C(12)-C(11)	122.8(2)	C(43)-C(42)-C(41)	113.81(11)
C(13)-C(12)-C(11)	117.7(2)	C(42)-C(43)-C(44)	113.90(11)
C(14)-C(13)-C(12)	122.4(2)	C(43)-C(44)-C(45)	113.48(11)
C(13)-C(14)-C(15)	116.6(2)		
C(13)-C(14)-C(26)	121.7(2)		
C(15)-C(14)-C(26)	121.7(2)		
C(16)-C(15)-C(14)	124.2(2)		
C(15)-C(16)-C(17)	117.2(2)		
C(15)-C(16)-C(30)	121.7(2)		
C(17)-C(16)-C(30)	121.2(2)		
O(2)-C(17)-C(12)	119.4(2)		
O(2)-C(17)-C(16)	120.6(2)		
C(12)-C(17)-C(16)	120.0(2)		
C(19)-C(18)-C(21)	108.06(19)		
C(19)-C(18)-C(20)	109.22(19)		
C(21)-C(18)-C(20)	106.75(19)		
C(19)-C(18)-C(2)	110.73(19)		
C(21)-C(18)-C(2)	111.86(19)		
C(20)-C(18)-C(2)	110.09(18)		
C(25)-C(22)-C(4)	112.25(19)		
C(25)-C(22)-C(24)	107.99(19)		
C(4)-C(22)-C(24)	109.44(17)		
C(25)-C(22)-C(23)	108.15(19)		
C(4)-C(22)-C(23)	110.00(19)		
C(24)-C(22)-C(23)	108.93(19)		
C(29B)-C(26)-C(28A)	132.5(5)		
C(29B)-C(26)-C(29A)	56.7(5)		
C(28A)-C(26)-C(29A)	112.9(4)		
C(29B)-C(26)-C(27B)	111.0(8)		
C(28A)-C(26)-C(27B)	36.6(7)		
C(29A)-C(26)-C(27B)	131.9(6)		
C(29B)-C(26)-C(14)	115.1(5)		
C(28A)-C(26)-C(14)	111.2(3)		
C(29A)-C(26)-C(14)	113.0(2)		
C(27B)-C(26)-C(14)	113.6(6)		
C(29B)-C(26)-C(27A)	47.7(6)		
C(28A)-C(26)-C(27A)	109.1(3)		
C(29A)-C(26)-C(27A)	103.7(3)		
C(27B)-C(26)-C(27A)	73.8(7)		
C(14)-C(26)-C(27A)	106.4(2)		
C(29B)-C(26)-C(28B)	106.6(7)		
C(28A)-C(26)-C(28B)	67.7(6)		
C(29A)-C(26)-C(28B)	52.4(5)		
C(27B)-C(26)-C(28B)	101.9(7)		
C(14)-C(26)-C(28B)	107.5(4)		
C(27A)-C(26)-C(28B)	144.4(4)		
C(31)-C(30)-C(32)	107.17(19)		
C(31)-C(30)-C(16)	110.21(19)		
C(32)-C(30)-C(16)	112.19(19)		
C(31)-C(30)-C(33)	110.10(19)		
C(32)-C(30)-C(33)	107.12(18)		
C(16)-C(30)-C(33)	109.96(18)		
C(35)-C(34)-Ti(1)	71.52(12)		
C(34)-C(35)-Ti(1)	68.89(12)		

Table 5. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^4$) for IAT87 (CCDC 822994). The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
Ti(1)	310(3)	256(2)	226(2)	3(2)	113(2)	44(2)
O(1)	249(10)	283(10)	201(8)	7(7)	80(7)	71(8)
O(2)	265(10)	290(10)	235(9)	47(8)	95(8)	50(8)
N(1)	226(12)	251(12)	189(10)	-21(9)	34(9)	25(10)
N(2)	292(14)	224(12)	378(13)	-12(10)	32(10)	91(10)
C(1)	237(15)	239(15)	199(13)	-53(11)	46(12)	27(12)
C(2)	281(16)	273(16)	193(13)	-25(11)	77(11)	18(12)
C(3)	316(16)	294(15)	189(13)	-24(11)	97(12)	-14(13)
C(4)	257(16)	288(15)	162(13)	-58(11)	34(11)	14(12)
C(5)	275(16)	246(15)	170(13)	-29(11)	21(11)	51(12)
C(6)	282(16)	223(14)	148(12)	-31(11)	52(11)	6(12)
C(7)	197(15)	273(15)	161(12)	-42(11)	-10(11)	46(12)
C(8)	313(16)	291(16)	217(13)	-22(12)	65(11)	85(13)
C(9)	327(16)	213(14)	259(14)	24(12)	49(12)	62(12)
C(10)	309(16)	296(16)	207(13)	54(12)	48(11)	20(13)
C(11)	214(15)	290(15)	196(13)	13(12)	28(11)	52(12)
C(12)	243(15)	272(15)	224(13)	9(12)	68(12)	28(12)
C(13)	284(16)	343(16)	241(14)	27(12)	56(12)	74(12)
C(14)	310(17)	372(17)	237(14)	4(13)	75(12)	31(13)
C(15)	327(17)	326(16)	261(14)	-44(12)	141(12)	-15(13)
C(16)	258(15)	259(15)	282(14)	-8(12)	109(12)	-25(12)
C(17)	238(15)	277(15)	212(13)	14(12)	68(12)	-19(12)
C(18)	302(17)	315(16)	236(14)	61(12)	88(12)	130(13)
C(19)	408(17)	444(18)	254(14)	41(13)	5(12)	157(15)
C(20)	457(18)	284(16)	366(15)	57(13)	133(13)	51(13)
C(21)	530(19)	550(20)	327(15)	215(14)	198(14)	251(15)
C(22)	217(15)	392(17)	215(13)	-33(13)	39(12)	31(13)
C(23)	299(18)	660(20)	471(18)	-130(16)	65(14)	-36(15)
C(24)	323(17)	517(19)	339(16)	-3(14)	143(13)	85(14)
C(25)	298(17)	790(20)	329(16)	18(15)	81(13)	194(15)
C(26)	480(20)	510(20)	195(15)	64(14)	98(14)	148(17)
C(27A)	660(40)	830(40)	260(30)	80(30)	-130(30)	10(30)
C(28A)	1040(60)	510(30)	330(30)	220(30)	330(40)	180(50)
C(29A)	710(40)	990(60)	270(20)	130(30)	200(30)	380(40)
C(27B)	1090(150)	1700(200)	70(50)	230(90)	10(100)	740(160)
C(28B)	680(90)	1290(160)	430(70)	520(90)	-110(60)	-400(90)
C(29B)	850(140)	960(110)	260(60)	150(70)	0(70)	-240(90)
C(30)	287(16)	274(15)	323(15)	11(12)	131(13)	39(13)
C(31)	268(16)	423(17)	467(16)	92(15)	87(13)	78(14)
C(32)	451(19)	449(18)	495(18)	75(15)	280(15)	149(14)
C(33)	473(18)	334(17)	431(16)	-27(13)	227(14)	91(14)
C(34)	363(17)	394(17)	302(15)	-95(13)	103(13)	-66(13)
C(35)	499(19)	251(15)	317(15)	-34(12)	196(13)	-22(13)
C(36)	570(20)	540(20)	407(16)	130(15)	174(14)	229(16)
C(37)	425(19)	520(20)	444(18)	111(15)	-25(14)	60(16)

C(41)	940(100)	580(80)	8700(400)	1150(170)	-530(180)	-70(80)
C(42)	2070(150)	2720(170)	1530(110)	-240(110)	280(110)	1720(110)
C(43)	1430(100)	2360(140)	1190(90)	990(100)	-370(80)	-890(100)
C(44)	1160(90)	2040(140)	1090(80)	960(90)	700(80)	-10(90)
C(45)	1050(80)	620(70)	1650(110)	410(60)	-690(70)	-10(60)
